

Polysaccharides

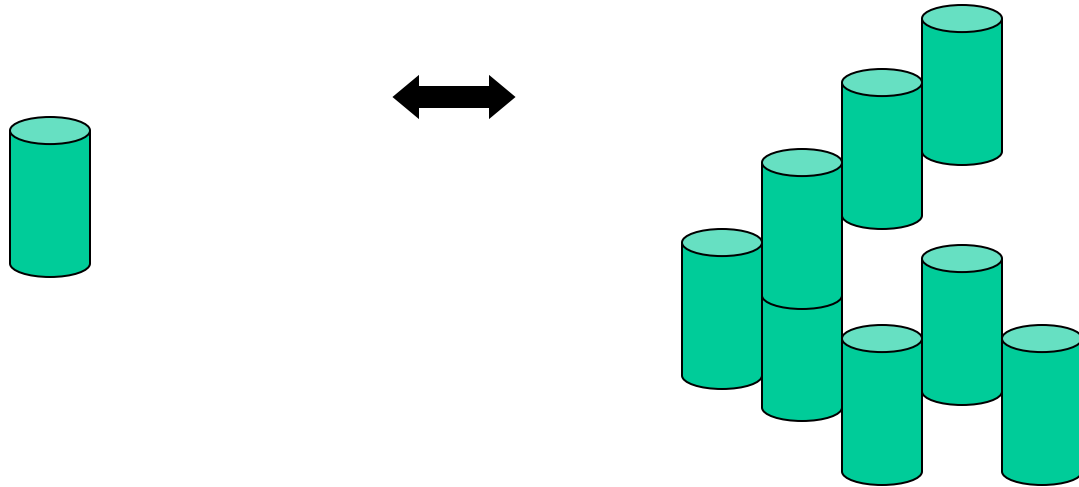
FDSC400

Sources of Polysaccharide

- Microbial fermentation
- Higher plants
 - seeds,
 - tree extrudates,
 - marine plants,
- Chemical modification of other polymers

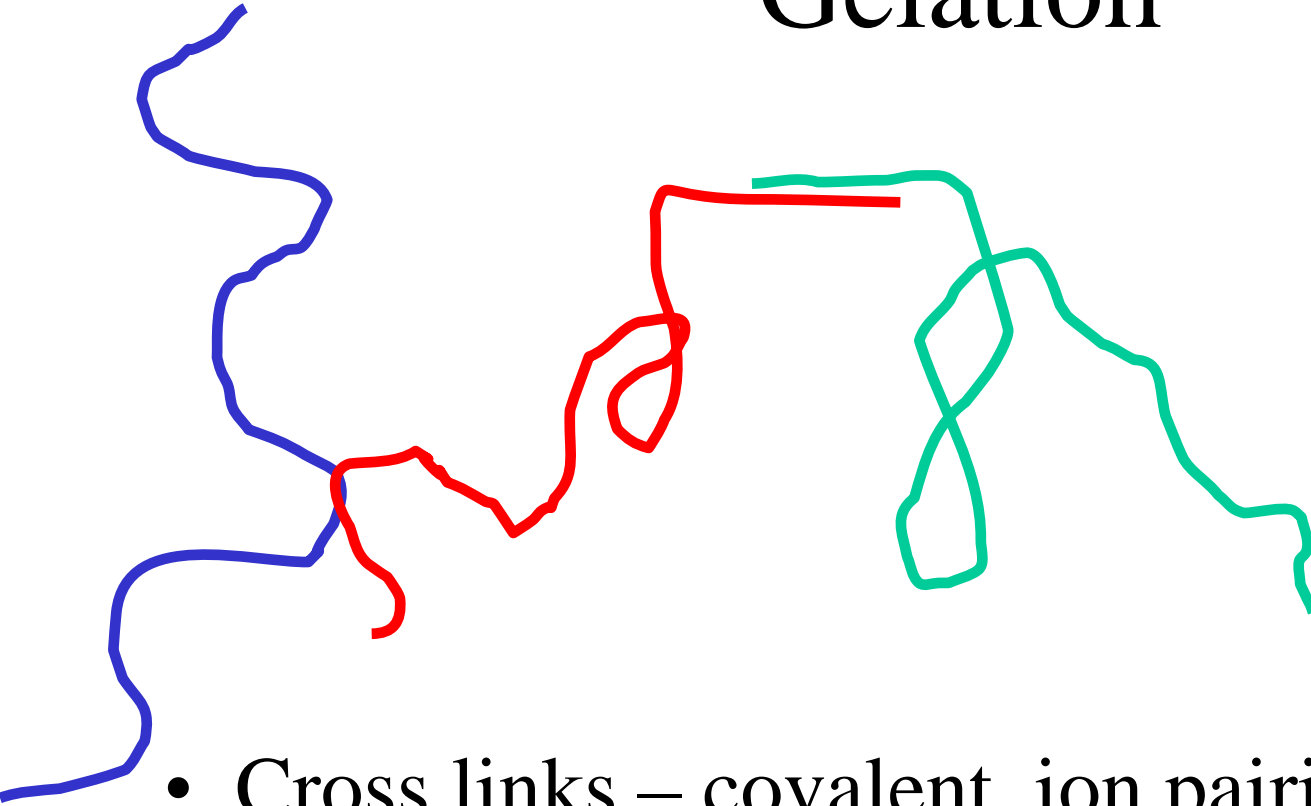
No calorific value; fiber

Solubility



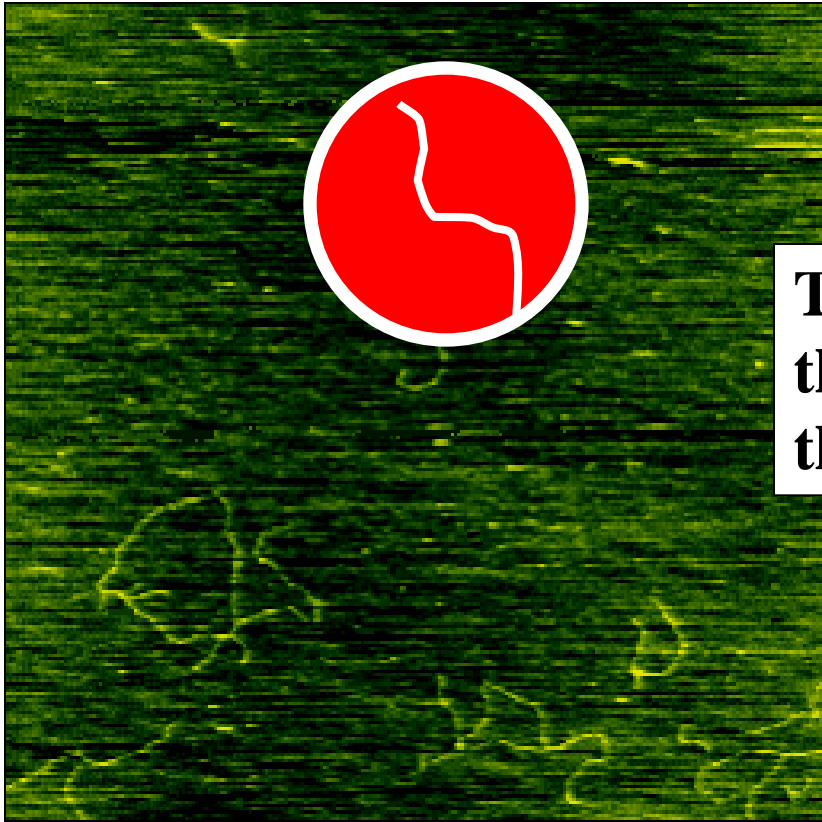
- Strong interaction with water = solubility
- Strong, extended interaction with polymer = insolubility
- Local, limited interaction with polymer = gelation

Gelation



- Cross links – covalent, ion pairing, co-crystallization

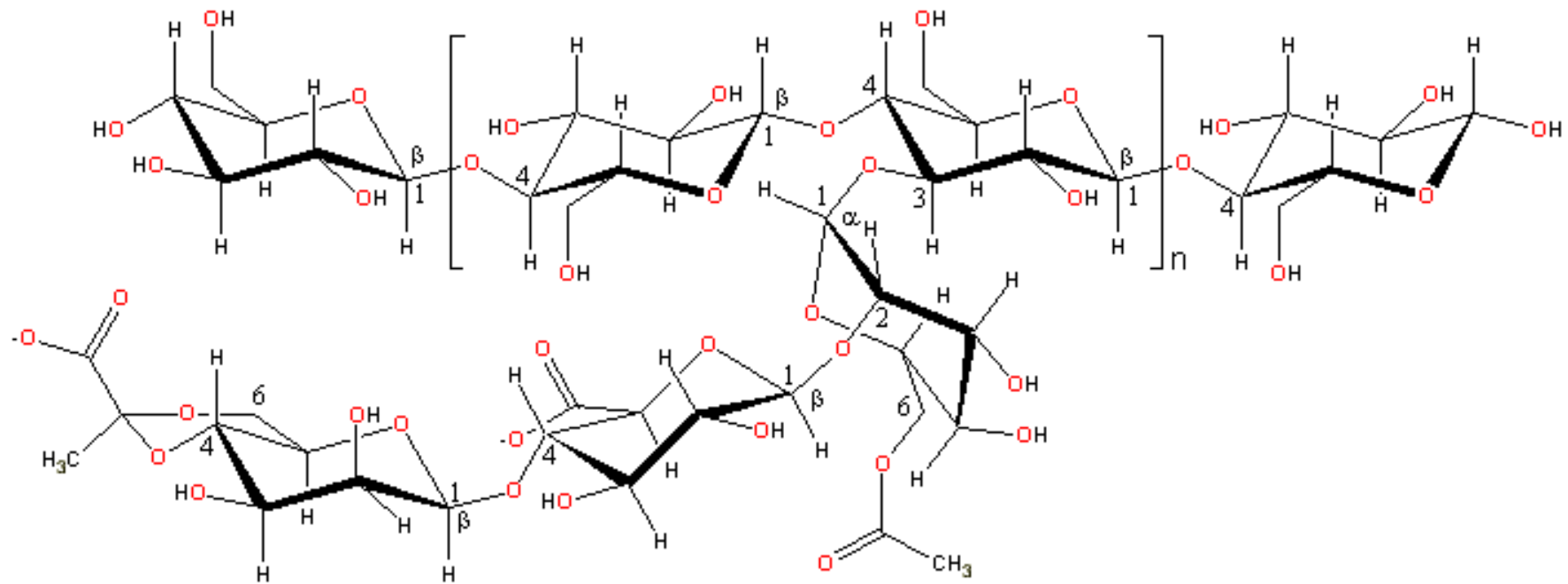
Viscosity



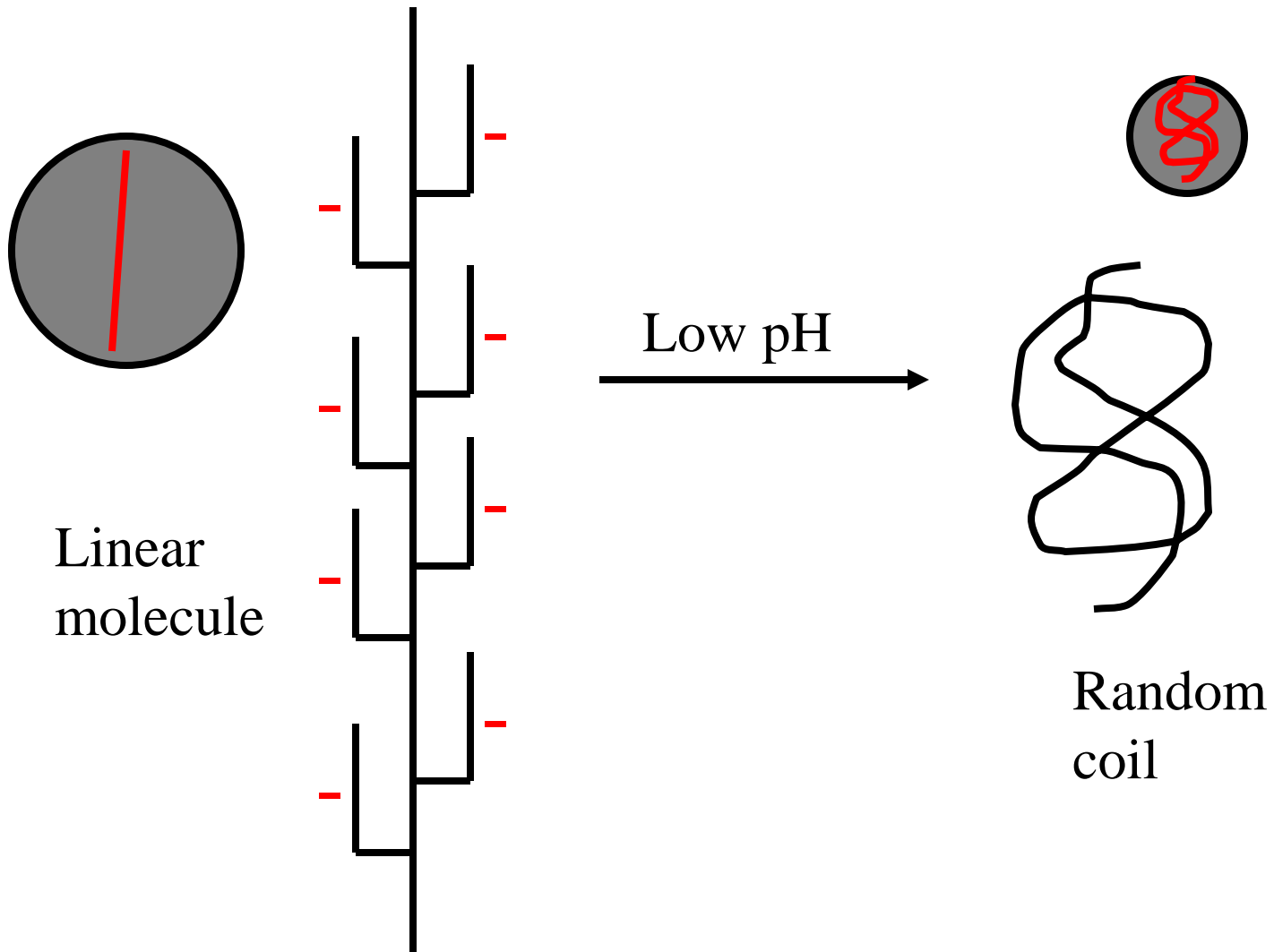
**The radius of gyration is
the sphere swept out by
the polymer**

Xanthan gum

- **Source**: Product of bacteria *Xanthomonas campestris*
- **Structure**: cellulose-like backbone (β -1,4-polyglucose) with trisaccharide branches (stubs) on alternate monomers on the backbone carrying carboxylic acid residue
- **Functional Properties**: Water soluble, viscous, non-gelling. Viscosity is only slightly temperature dependant

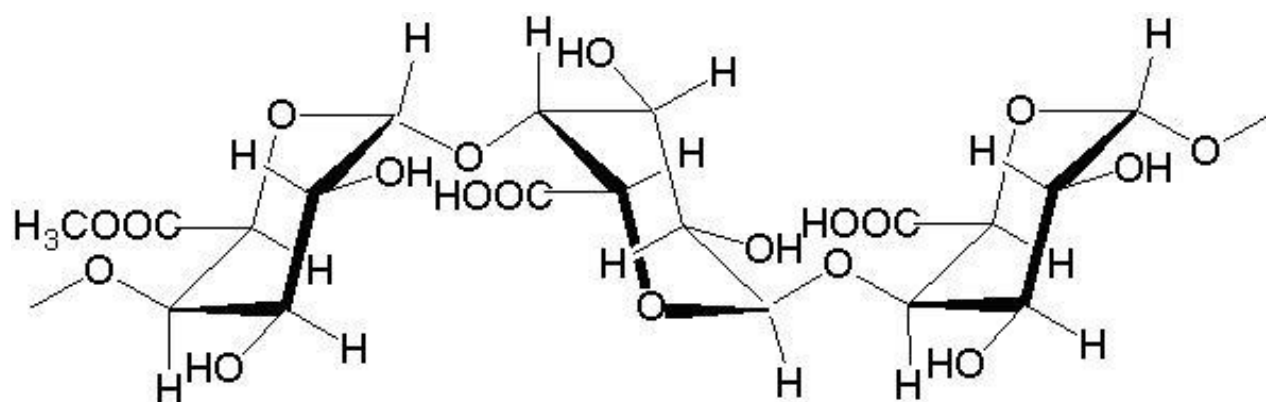
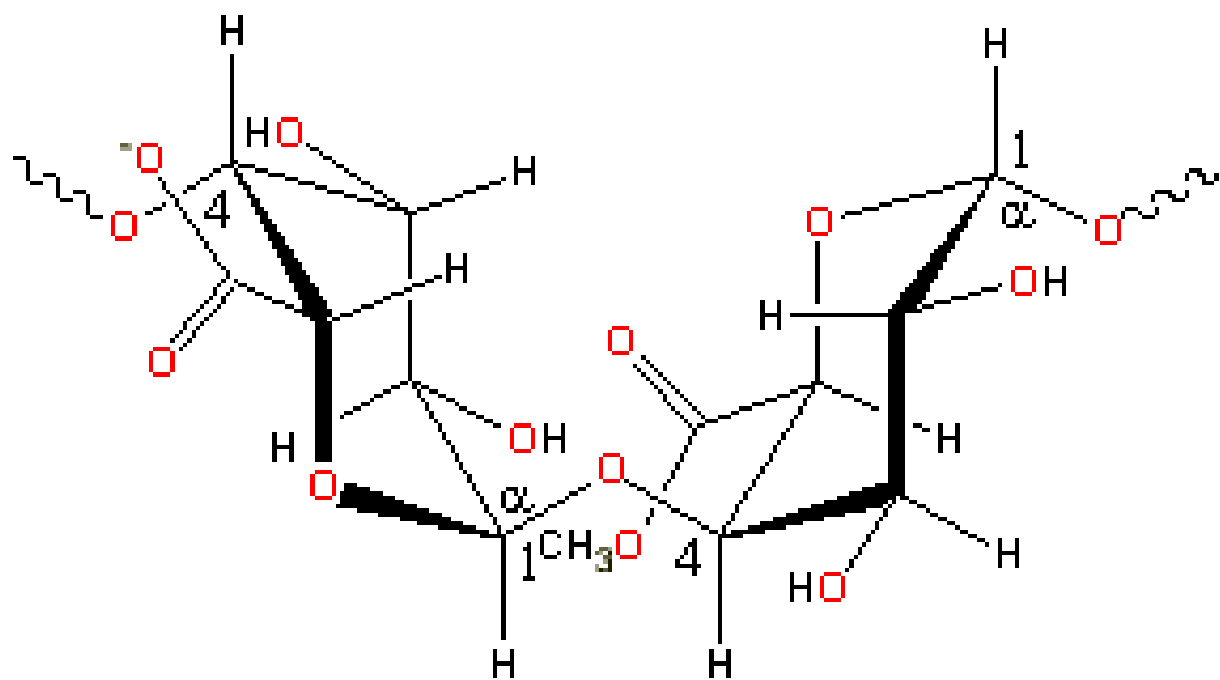


Xanthan: Structure-function

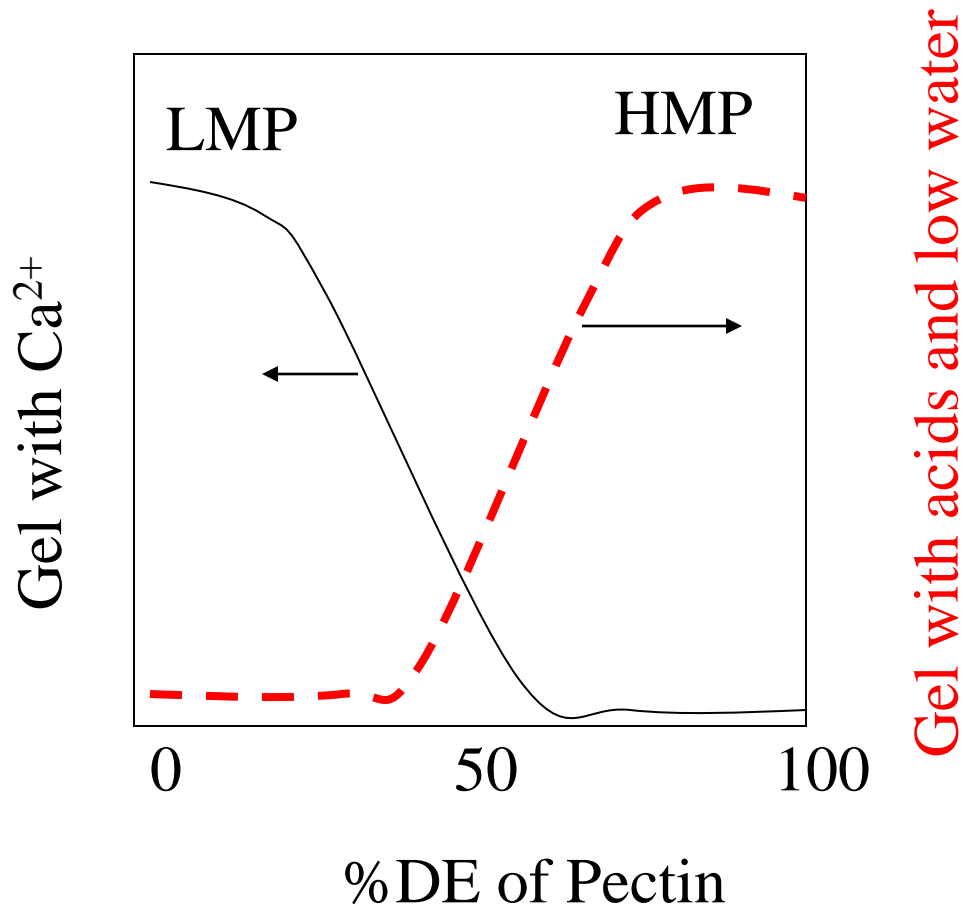


Pectin

- **Source**: Cell walls of higher plants (citrus rind)
- **Structure**: Largely a linear polymer of polygalacturonic acid with varying degrees of methyl esterification. (Also some branches – HAIRY REGIONS)
 - >50% esterified is a high methoxy (HM) pectin
 - <50% esterified is a low methoxy (LM) pectin
- **Functional Properties**: High methoxy pectin will gel in the presence of acid and high sugar concentrations. Low methoxy pectin will gel in the presence of calcium.

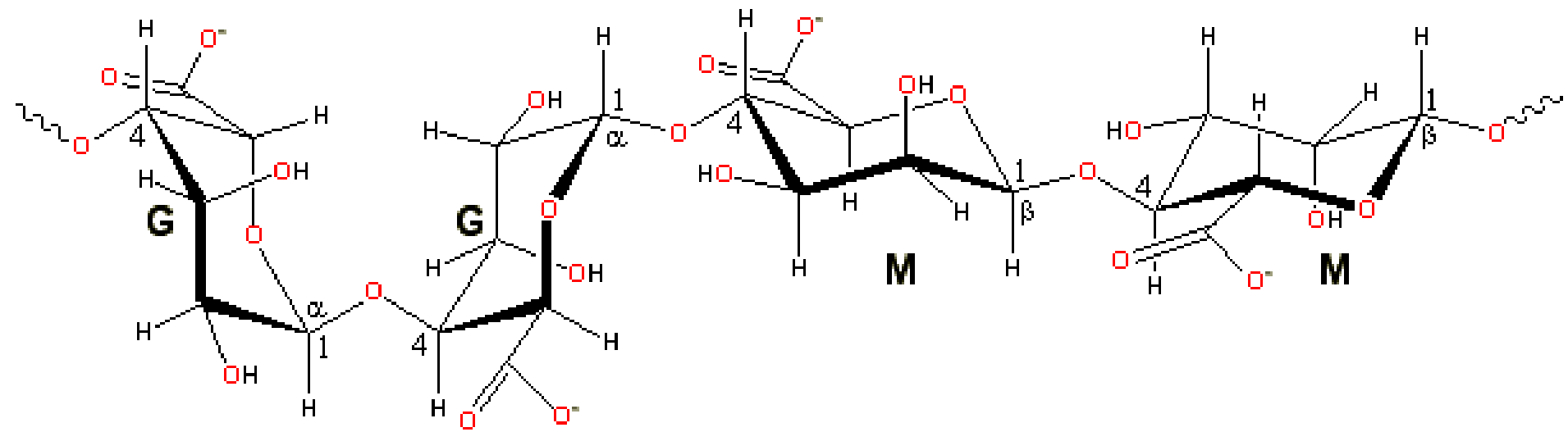


High and Low Methoxy Pectin

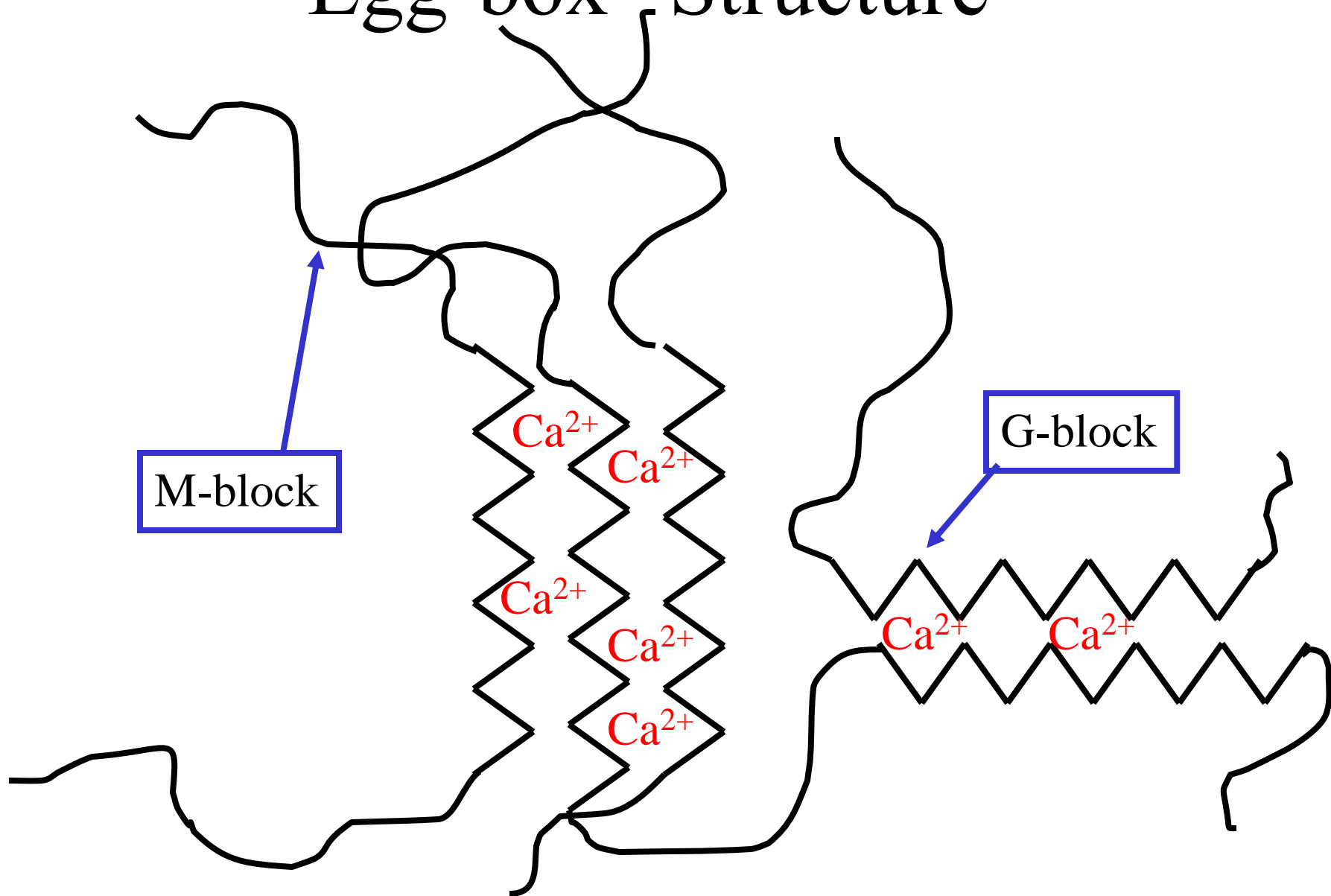


Algin

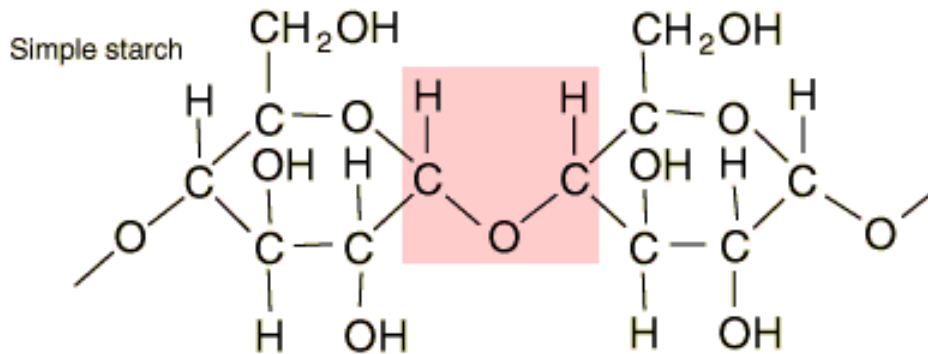
- **Source**: Seaweed extract
- **Structure**: linear polysaccharide containing two types of residue (i.e., a co-polymer): β -D-mannopyranosyluronic acid and (M) α -L-gulopyrasonic acid (G)
- **Functional Properties**: Viscous in aqueous solution, gels in the presence of Ca^{2+} (or low pH). Gels are temp stable
- PGA (propylene glycol alginate)



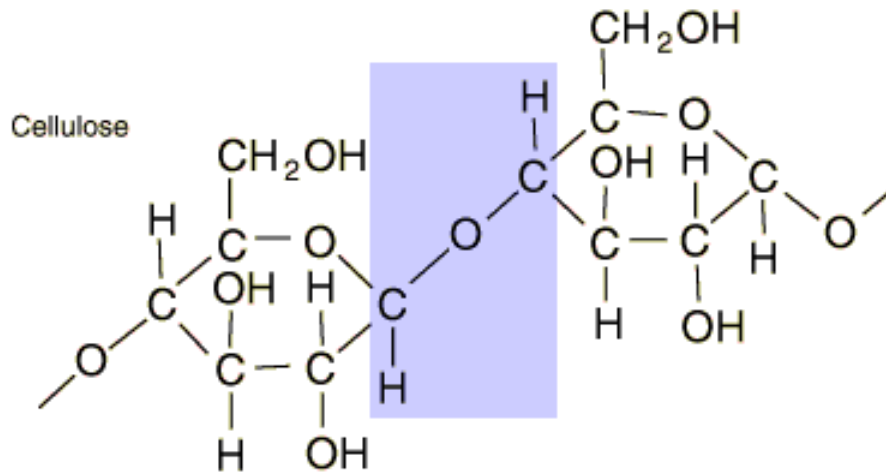
“Egg-box” Structure



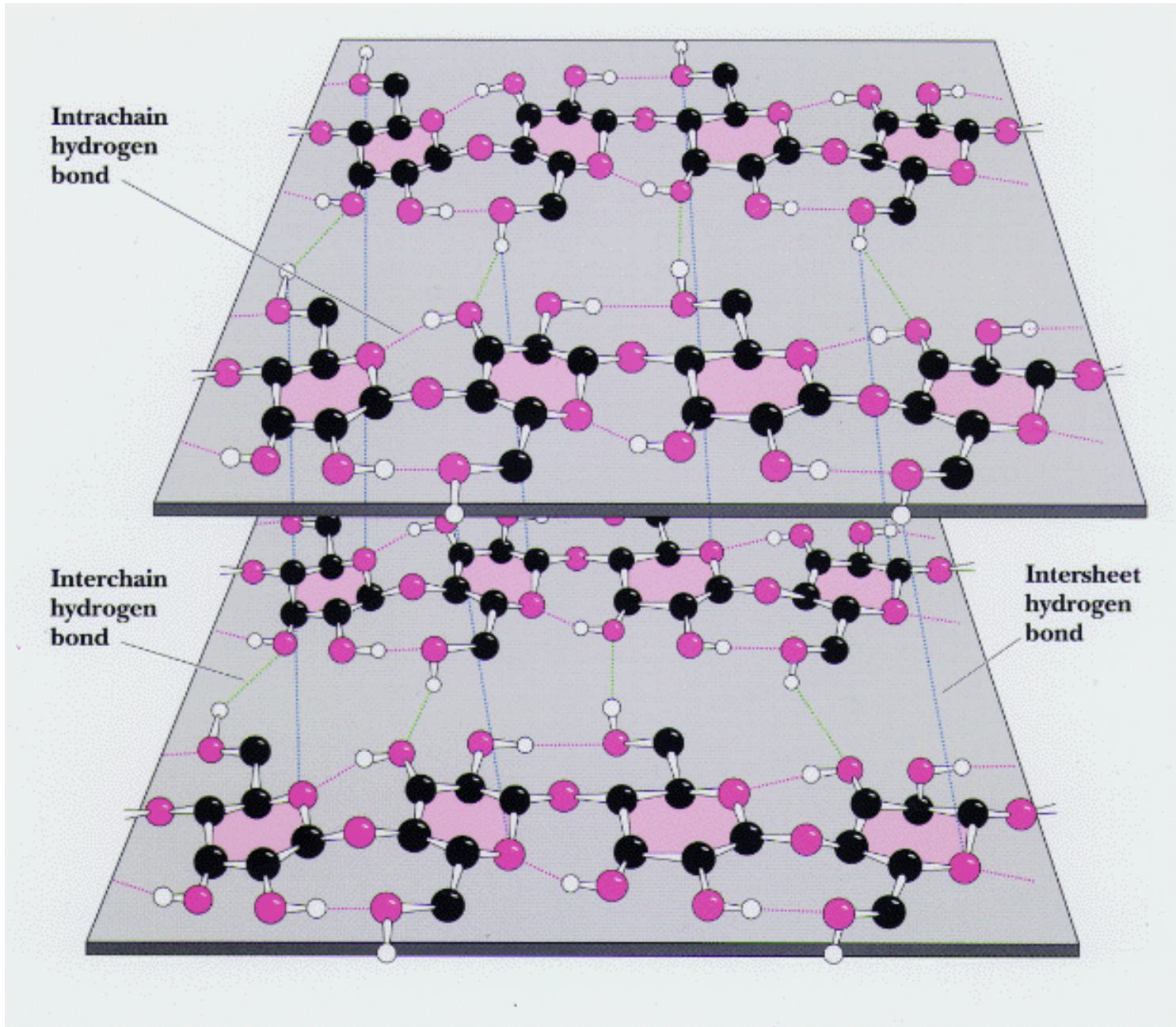
Cellulose Gums



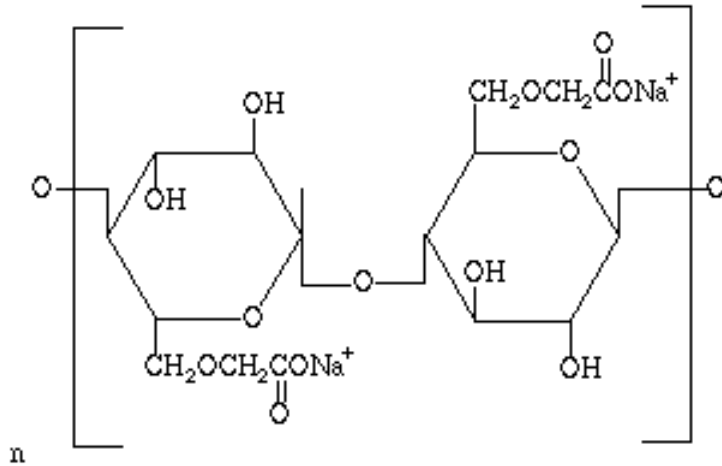
α 1-4



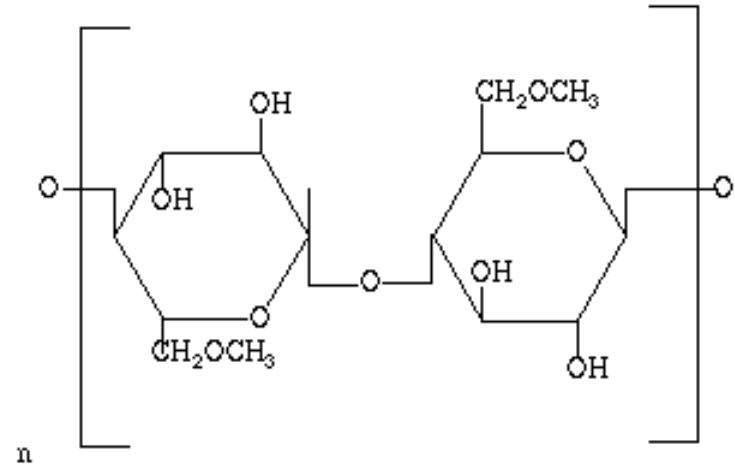
β 1-4



Carboxymethylcellulose



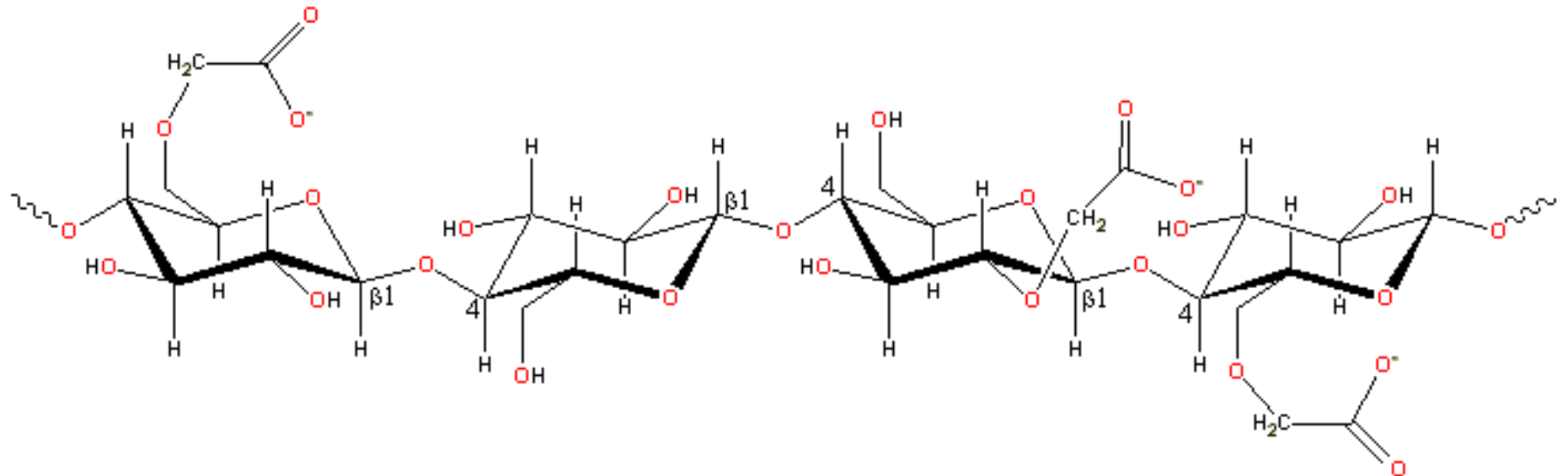
Methylcellulose



(may gel at high T)

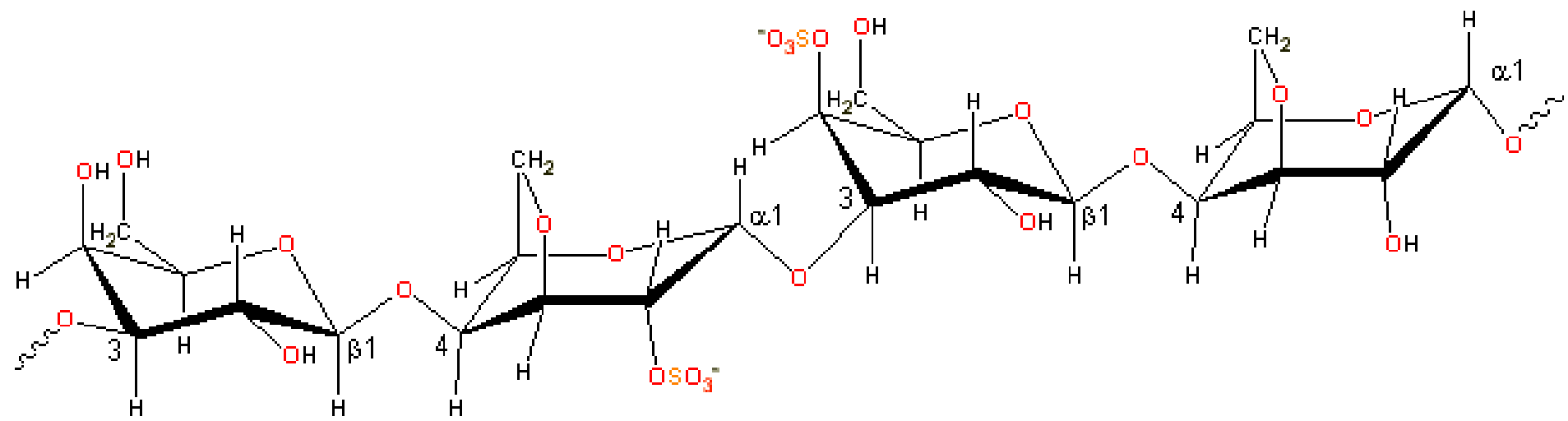
Carboxymethyl cellulose

- $-\text{CH}_2\text{COO}^-$
- High viscosity, non-gelling



Carageenan

- **Source**: Seaweed gum
- **Structure**: Linear D-galactopyranosyl chain with alternating 1,3 and 1,4 links. Some residues have one or two sulfate ester residues. Three broad types of repeating structure (ι , κ , and λ carageenan)
- **Functional Properties**: pH independent thickening. Double helix formation in κ or ι carageenan can lead to gelation.
 - κ -carageenan in dairy foods



Gum Arabic

- Extrudate gum of the acacia tree
- Expensive – hard to source
- Low viscosity, non-gelling
- Complexed with a glycoprotein -surface active