

## SPEKTROKEMIJSKA ANALIZA

- SPEKTROSKOPIJA
- SPEKTROMetriJA

### OPTIČNA SPEKTROKEMIJSKA ANALIZA

- ATOMSKA SPEKTROSKOPIJA
- MOLEKULARNA SPEKTROSKOPIJA

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*"Spectrochemical Analysis"*: Methods of chemical analysis that depend upon the measurement of the wavelength and the intensity of electromagnetic radiation...  
*Encyclopedia Britannica*



1670 Lom svetlobe, spekter

*Isaak Newton*  
(1643-1727)



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### SPEKTROSKOPIJA: Odkritja novih elementov

*Robert Wilhelm Bunsen* (1824 – 1887)

*Gustav Robert Kirchhoff* (1811 – 1899)



1860 Cs, Rb  
1861 Tl  
1863 In  
1868 He  
1875 Ga ...

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## KEMIJSKA ANALIZA

- ANALIZA GLAVNIH KOMONENT
- ANALIZA SLEDOV

MAKRO ANALIZA, SEMIMIKRO ANALIZA,  
MIKRO-ANALIZA, ULTRAMIKRO ANALIZA  
ULTRA-ANALIZA SLEDOV

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## SPEKTROKEMIJSKA ANALIZA

- EMISIJA (FLUORESCENCA)
- ABSORPCIJA

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## SPEKTROKEMIJSKA ANALIZA

KALIBRACIJSKA FUNKCIJA

$$S = f(C_a, \lambda, X_i)$$

ANALITSKA FUNKCIJA

$$C_a = g(S)$$

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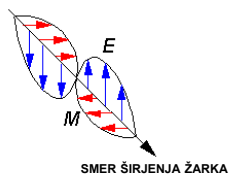
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## Elektromagnetno valovanje



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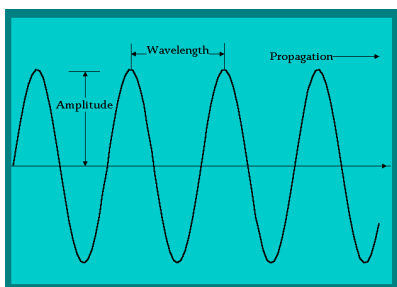
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## Elektromagnetno valovanje



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## Osnovne zveze

$$E = h \cdot \nu = h \cdot \frac{c}{\lambda}$$

- **E.....energija v J**
- **$\nu$ .....frekvenca v Hz,  $s^{-1}$**
- **$\lambda$ .....valovna dolžina**
- **h.....Planckova konstanta,  $6,63 \cdot 10^{-34}$  Js**
- **c.....hitrost svetlobe,  $3,00 \cdot 10^8$   $ms^{-1}$**

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## Enote

### Enote za energijo

- J
- $\text{erg}=10^{-7} \text{ J}$
- $\text{eV}=1,6 \cdot 10^{-19} \text{ J}$

### Enote za valovno dolžino:

- $1\text{\AA}=10^{-10} \text{ m}$
- $1\text{nm}=10^{-9} \text{ m}$
- $\text{mm}=10^{-6} \text{ m}$
  
- $1 \text{ eV} \dots\dots 1240 \text{ nm}$

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Anders Jonas Angström  
(1814 – 1874)

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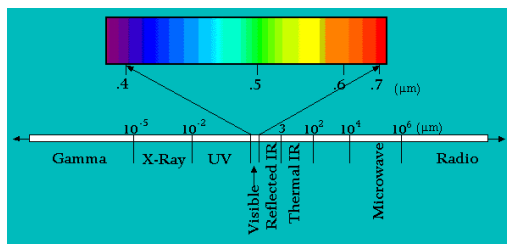
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## Spekter elektromagnetnega valovanja



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### Spekter vidne svetlobe



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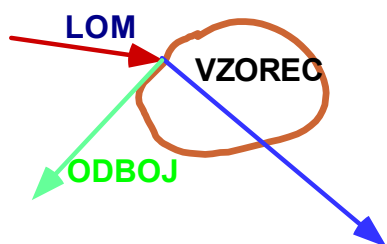
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### Interakcija med svetlobo in snovjo



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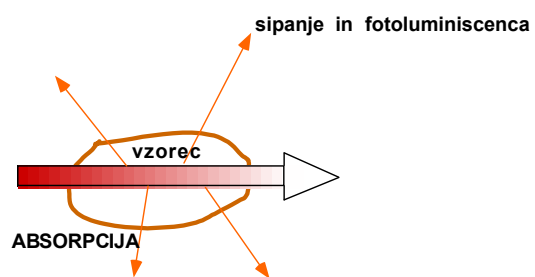
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### Absorpcija



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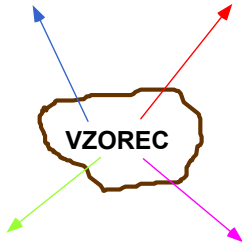
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## Emisija

### EMISIJA (FLUORESCENCA)



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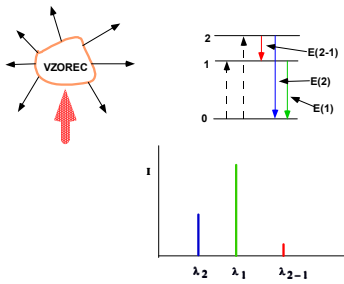
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## Emisija



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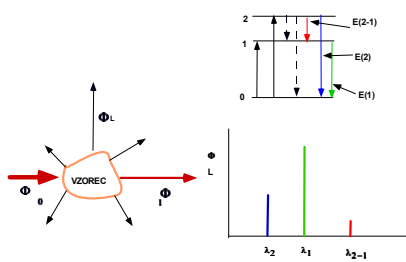
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## Absorpcija in fluorescenca



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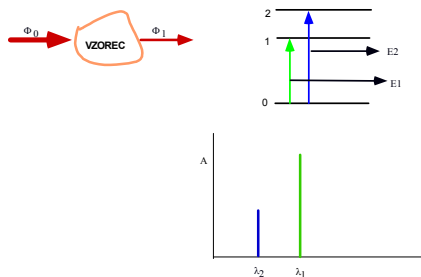
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## Absorpcija elektromagnetnega valovanja



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## Spektri

Kaj je spekter?

- Odvisnost merjenega signala (absorpcija, emisija) od valovne dolžine

Ločimo absorpcijske in emisijske spektre

- **Atomijski spektri** – črtasti (diskretni prehodi med energetskeimi stanji atomov)
- **Molekulski spektri** – zvezni (zvezni prehodi med energetskeimi stanji molekul)

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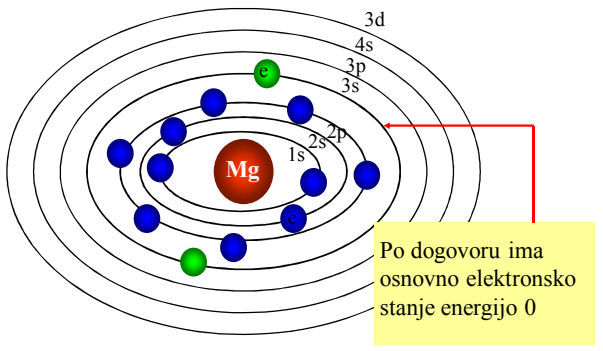
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## ATOMI: (primer:Elektronska konfiguracija Mg)



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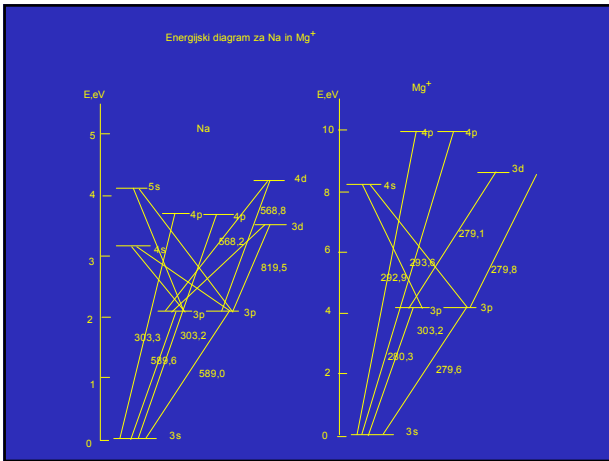
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**Elektronski prehodi (UV-VIS):**

**Atomi:**  
Prehodi zunanjih elektronov (črtasti spektri)

**Molekule:**  
Prehodi veznih ali neveznih elektronov (zvezni spektri)

$\pi \longrightarrow \pi^*$

$n \longrightarrow n^*$

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**Molekule: Absorpcija**

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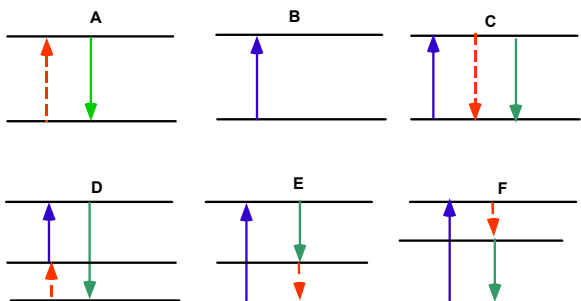
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Glavne vrste optičnih prehodov




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Shema aparature za merjenje emisije




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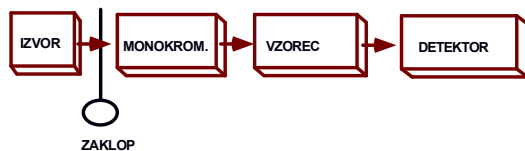
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Shema aparature za merjenje absorpcije




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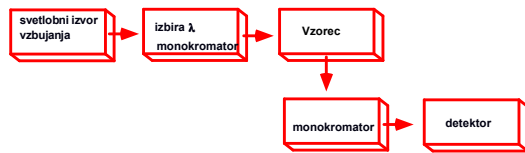
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Shema aparature za merjenje  
fluorescence



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