

Izosterija in bioizosterija

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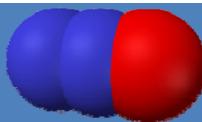
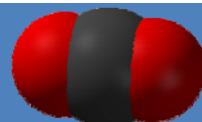
15. November 2012

Izosterija

Gr. Izo = enak, steros = trden, oprijemljiv, prostorski

1919 fizik Langmuir definira:

- Izosterične skupine (atomi, ioni, molekule) – enako število elektronov in enako št. atomov
- Izosterija = fizikalno-kemijska podobnost, zamenjava ne povzroči sprememb v fiz.-kem lastnostih

	N_2O 	CO_2 
Gostota (10 °C)	0,856	0,858
Ref. Indeks (16 °C)	1,193	1,190
Dielektrična konstanta	1,593	1,582
Topnost v etanolu (15°C)	3,250	3,130

Izosterija

1925 Grimmov “zakon” hidridnih zamjenjav:

- “Atoms anywhere up to four places in the periodic system before an inert gas change their properties by uniting with one to four hydrogen atoms, in such a manner that the resulting combinations behave like **pseudoatoms**, which are similar to elements in the groups one to four places respectively, to their right.”

	Group 4A	Group 5A	Group 6A	Group 7A	Group 8A	
Nº of e ⁻	6	7	8	9	10	11
	C	N	O	F	Ne	Na ⁺
	H ↗ CH	NH	OH	FH		
	H ↗ CH ₂	NH ₂	OH ₂	FH ₂ ⁺		
	H ↗ CH ₃	NH ₃	OH ₃ ⁺			
	H ↗ CH ₄	NH ₄ ⁺				

Izosterija

Grimmov "zakon" – ali drži?

	Pentan	Dietileter	Dietilamin
Vrelišče	36,1 °C	34,6 °C	55,5 °C
Viskoznost (20°C)	0,240 cP	0,224 cP	~0,35 cP
Ref. Indeks n_{20}/D	1,358	1,353	1,385
Gostota	0,626 g/cm ³	0,7134 g/cm ³	0,7074 g/cm ³

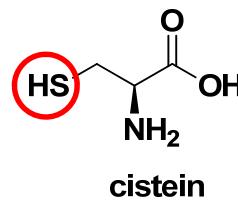
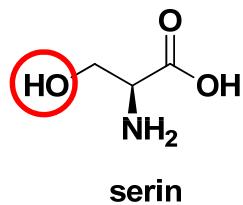
Izosterija

1932 Erlenmeyer razširi pojem:

- Izosteri so elementi, funkcionalne skupine, ioni in molekule z enakim številom valenčnih elektronov
- Elementi iste skupine so izosteri

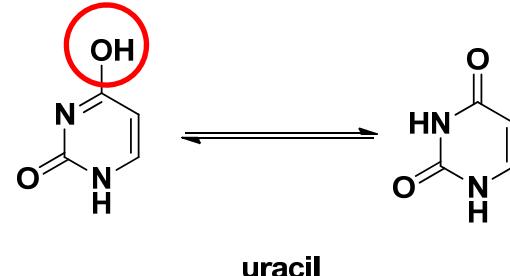
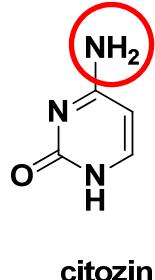
Št. Valenčnih el.	4	5	6	7	8
	N+	P	S	Cl	ClH
	P+	As	Se	Br	BrH
	S+	Sb	Te	I	IH
	As+		PH	SH	SH ₂
	Sb+			PH ₂	PH ₃

Izosterija v naravi



POSLEDICE?

- disulfidne vezi (proteini)
- nukleofilnost (aktivno mesto encimov)
- velikost
- lipofilnost



POSLEDICE

- AT/U (2 H-vezi)
- CG (3 H-vezi)

Izosteri

Izosteri = atomi, skupine, ioni, molekule

- **imajo podoben volumen,**
- **podobno obliko,**
- **enako število elektronov in njihovo razporeditev v zunanji lupini**

Bioizosteri

Friedman 1951:

Bioizosteri = izosteri z enakim (podobnim)
biološkim učinkom

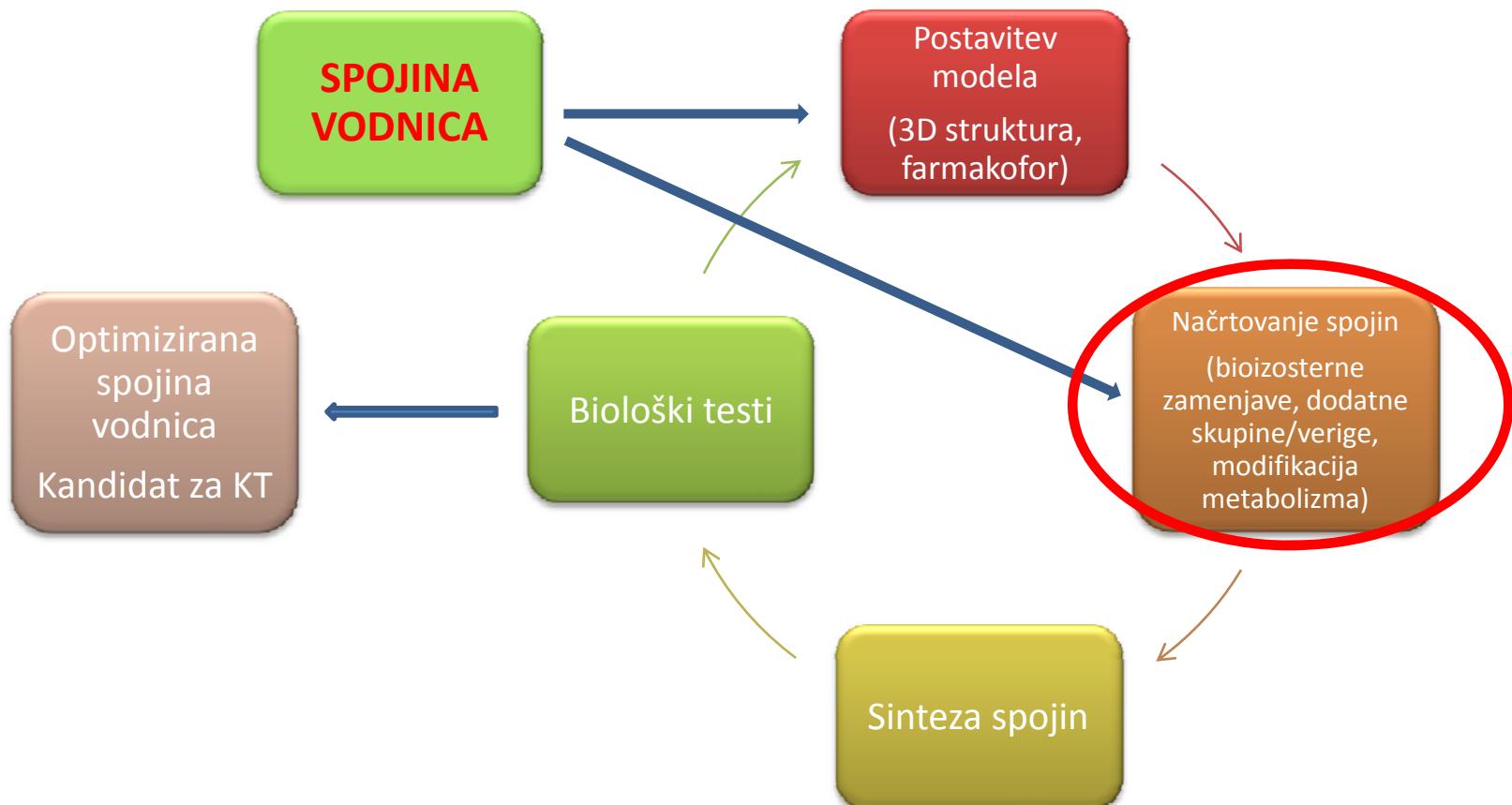
Burger 1990:

= atomi, skupine, ioni, molekule

- imajo podoben volumen,
- podobno obliko,
- podobno število elektronov in njihovo razporeditev v zunanji lupini,
- enako tarčo,
- agonisti/antagonisti.

} podobne
fiz-kem.
lastnosti

Optimizacija spojine vodnice – racionalno načrtovanje učinkovin



Čemu bioizosterne zamenjave?

Nove spojine

- “me too” princip – nove, patentibilne spojine

Modifikacije spojine vodnice – načrtovanje analogov

- Sprememba učinka
- Jakost
- Selektivnost
- Biol. uporabnost, boljše ADME lastnosti
- Zmanjšanje str. učinkov
- Kemijska stabilnost
- Stroški sinteze

Delitev bioizosterov

Klasični

- Ustrezajo osnovnim pogojem izosterov (Langmuir, Grimm)
- Delitev na podskupine

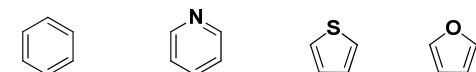
MONOVALENTNI BIOIZOSTERI

F, H
OH, NH
F, OH, NH ali CH₃ namesto H
SH, OH
Cl, Br, CF₃

DVOVALENTNI BIOIZOSTERI (vpliv na 2 vezi!)



OBROČNI EKVIVALENTI



TRIVALENTNI ATOMI ALI SKUPINE (vpliv na 3 vezi)



TETRASUBSTITUIRANI ATOMI ALI SKUPINE (vpliv na 4 vezi!)



Neklasični

Monovalentni bioizosteri

Vodik – fluor

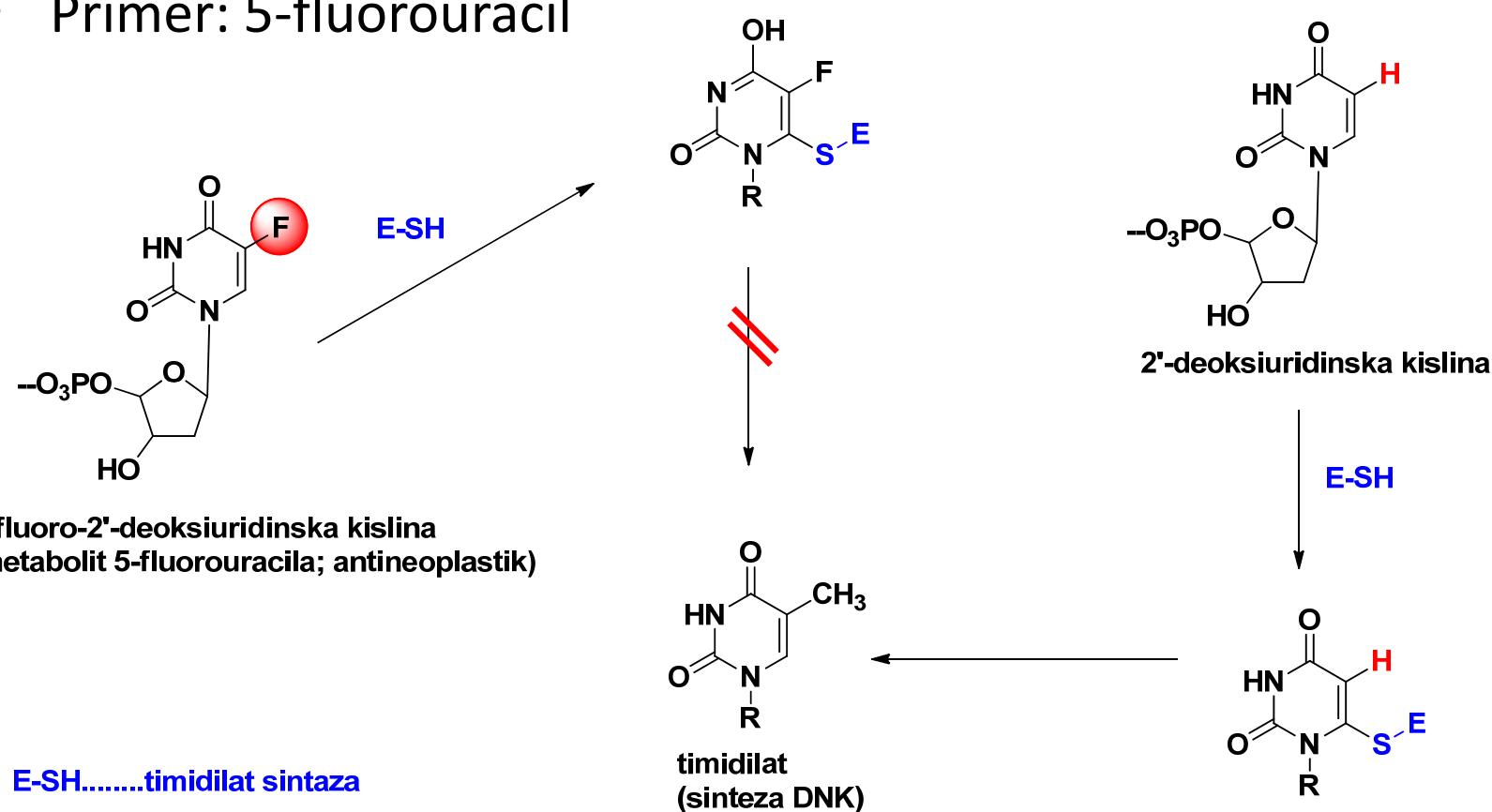
- Podobna radija ($1,2 - 1,35 \text{ \AA}$), ohranjena lipofilnost
- Povećana elektronegativnost
- Metabolična obstrukcija

	H	F	Cl	CH ₃	CF ₃
VdW radius	1.2	1.35	1.80	2	2
MR	1.03	0.92	6.03	5.65	5.02
induktivni efekt	-	3.08	2.68	0.00	2.85
π	0	0.14	0.71	0.56	0.88

Monovalentni bioizosteri

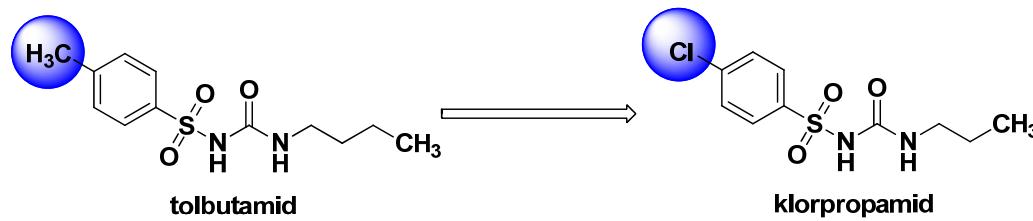
Vodik – fluor

- Primer: 5-fluorouracil



Monovalentni bioizosteri

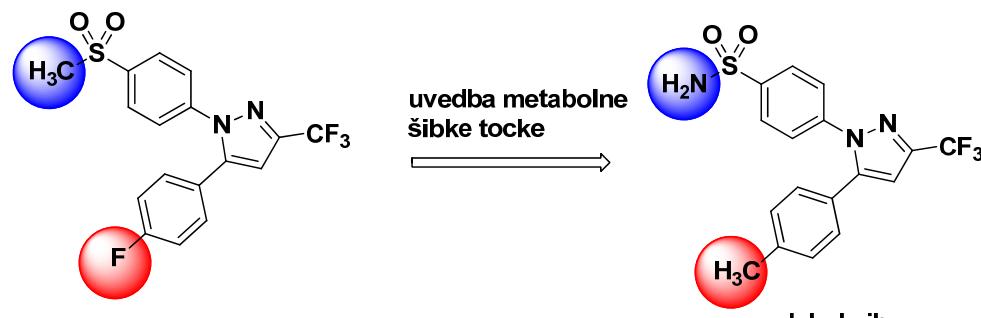
-F, -OH, -NH₂, -CH₃



$t_{1/2} = 4-5 \text{ h}$

METABOLNA STABILNOST
-daljši razpolovni čas

$t_{1/2} = 36 \text{ h}$



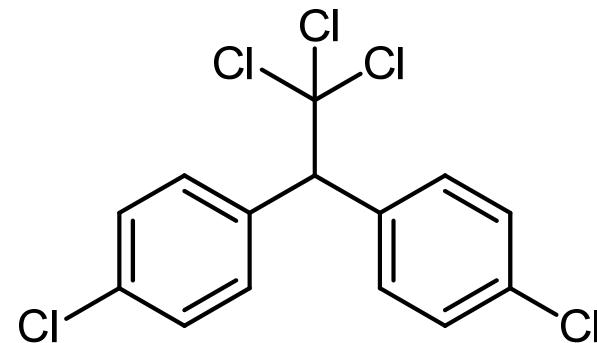
$t_{1/2} = 211 \text{ h}$
selektivnost (COX-2/COX-1) > 1000

$t_{1/2} = 8-12 \text{ h}$
selektivnost (COX-2/COX-1) = 375

Monovalentni bioizosteri

-Cl, -OH, -CH₃

- Grimmov "zakon" + Erlenmayerjeva razširitev
- DDT = **diklorodifeniltrikloroetan**



Monovalentni bioizosteri

-OH, -NH₂ skupina

- Podobna velikost
- Polarnost
- Akceptorji/donorji H-vezi
- Folna kislina/aminopterin/metotreksat

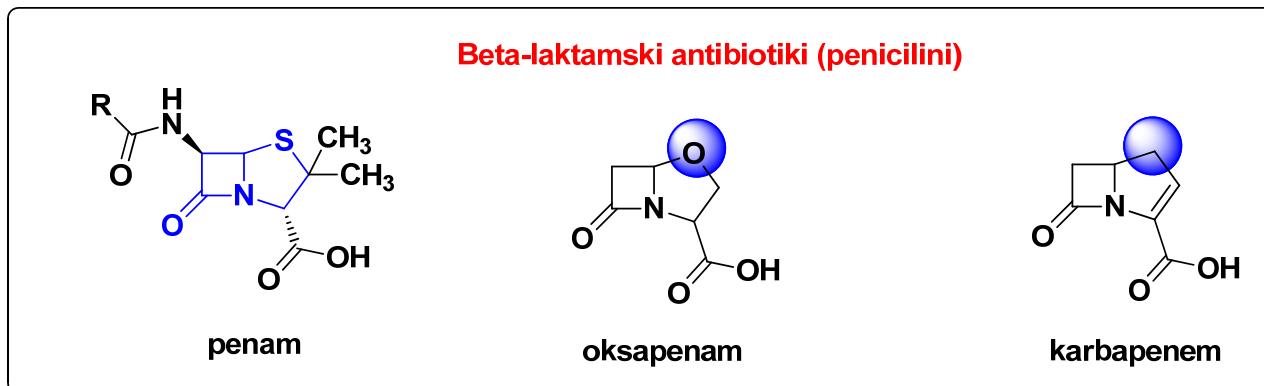
-OH, -SH skupina

- Podobna polarnost
- Različna tvorba H-vezi
- Gvanin/tiogvanin

Dvovalentni bioizosteri

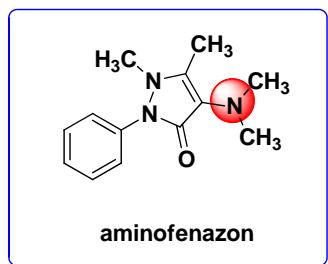
-CH₂-, -NH-, -O-, -S-

- β-laktamski antibiotiki

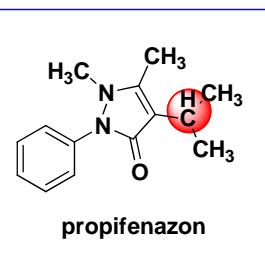
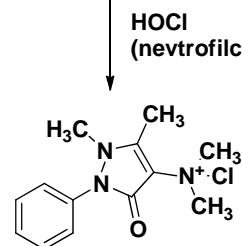


Trivalentni bioizosteri

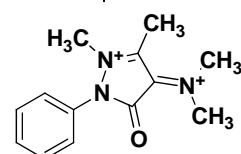
-CH=, -N=, -P=, -As=



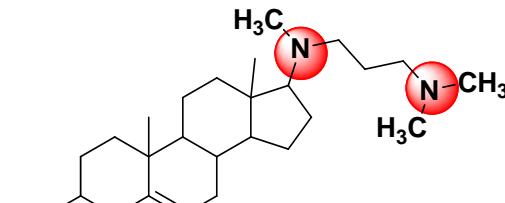
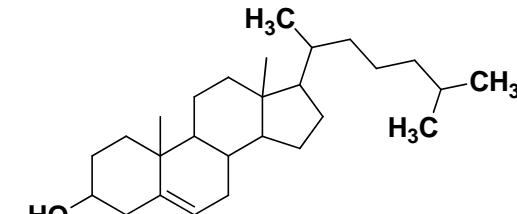
-antipiretika



NI TOKSIČNIH
METABOLITOVI!



TOKSIČNI
METABOLITI!!

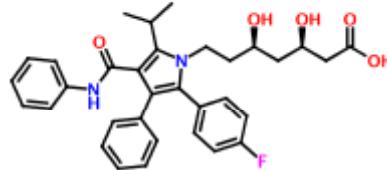


-inhibitor biosinteze holesterola

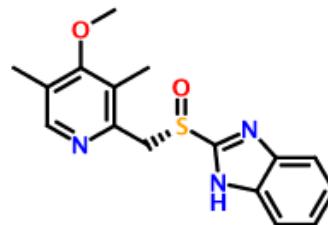
Obročni ekvivalenti

- Najpogosteje uporabljan koncept bioizosterije
- 8/10 najbolje prodajanih zdravil (2006):

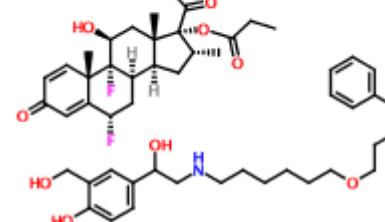
atorvastatin



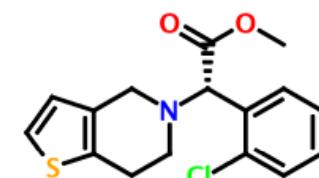
esomeprazol



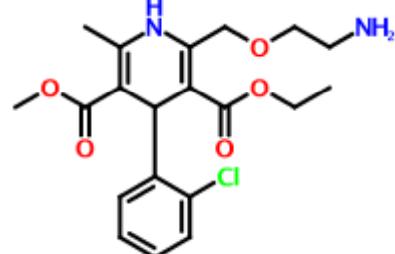
Flutikazol/
salmeterol



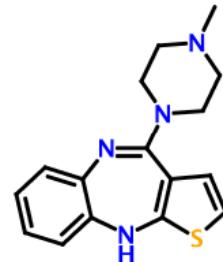
klopidogrel



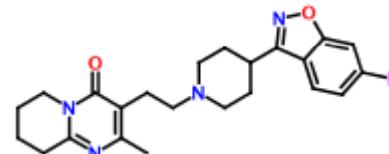
amlodopin



olanzapin



risperidon



venlafaksin



Obročni ekvivalenti

Tiofen – bezen (Meyer 1883)

MW = 78.11

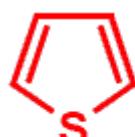
mp. = 5.5°C

bp. = 80.15°C

Log P = 2.13

MR = 26.4

d = 0.879



MW = 84.14

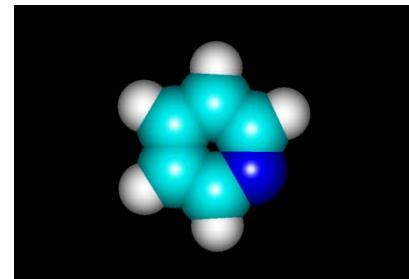
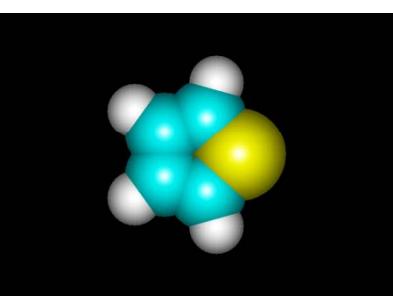
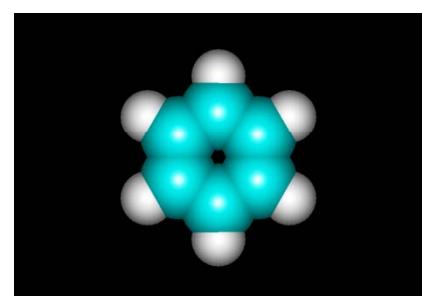
mp. = -38°C

bp. = 84°C

Log P = 1.81

MR = 25.0

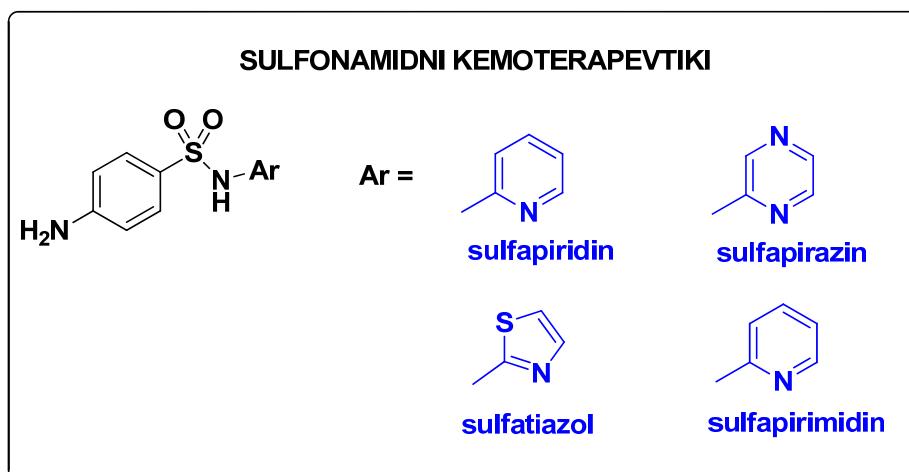
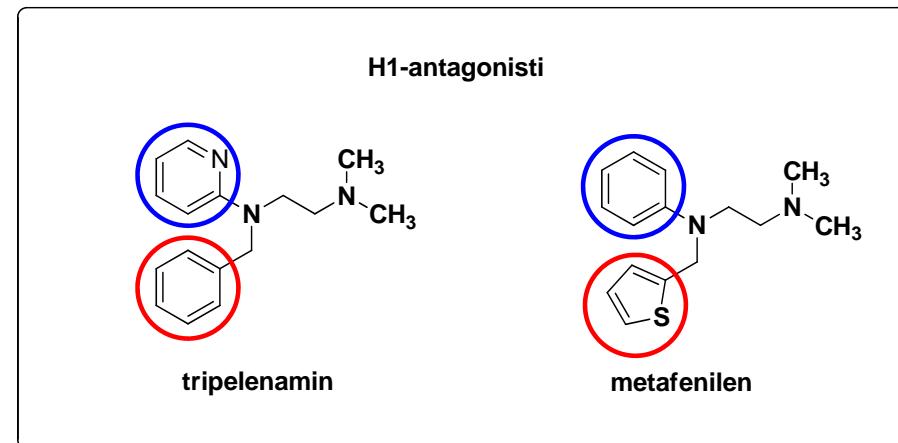
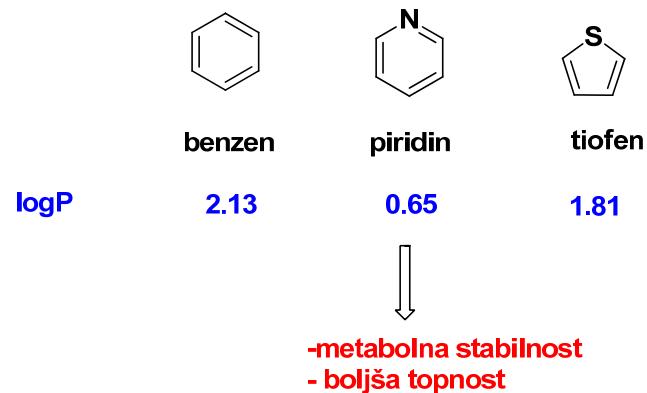
d = 1.057



Obročni ekvivalenti

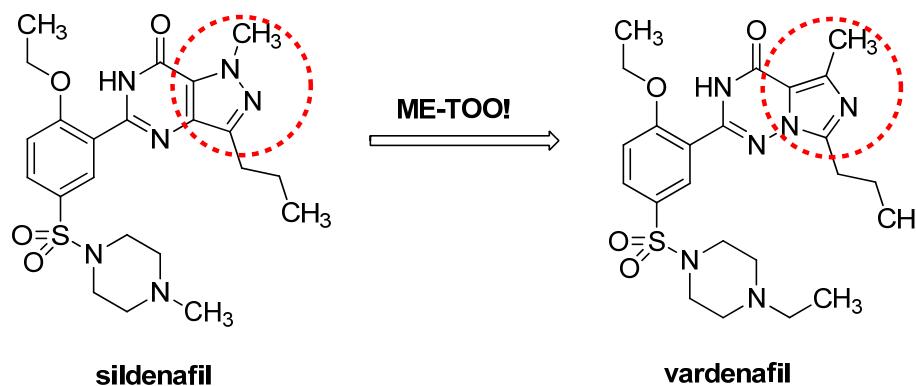
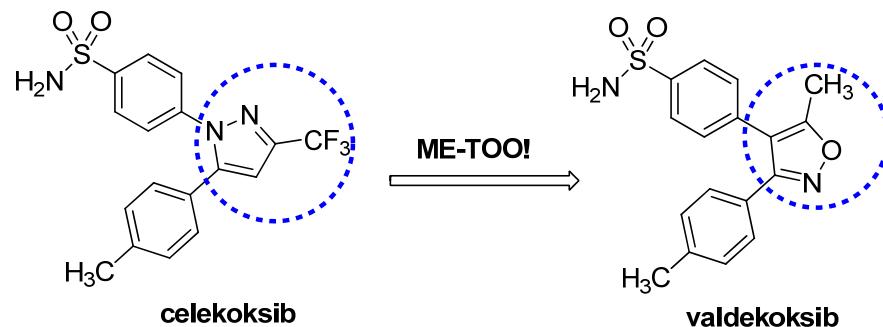
Klonidin – tiamenidin (α_2 -agonista)

Obročni ekvivalenti



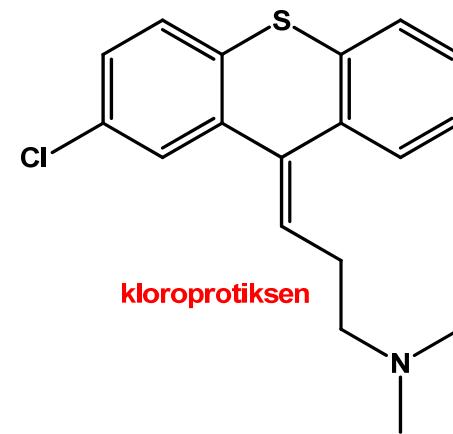
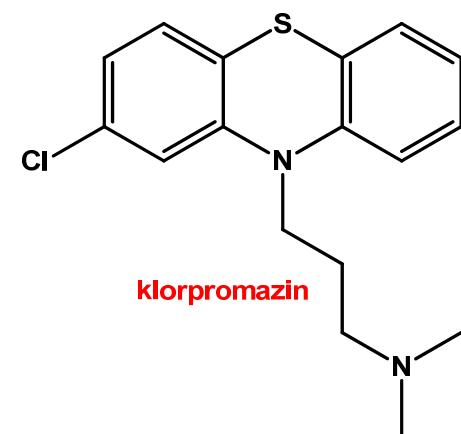
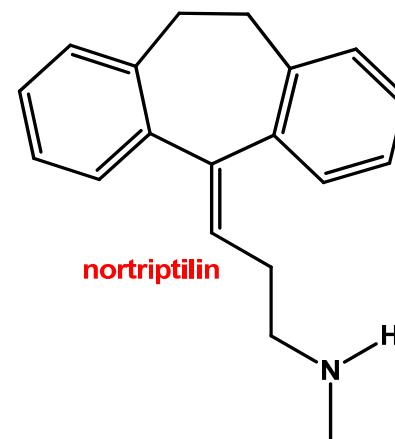
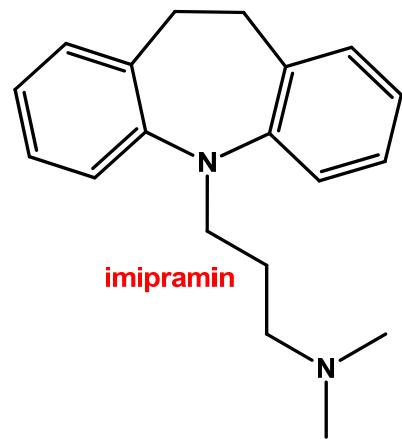
Obročni ekvivalenti

- “Me-too”



Obročni ekvivalenti

- Triciklični antidepresivi/nevroleptiki



Neklasični bioizosteri

Neklasični bioizosteri - klasični

- Podobna prostorska razporeditev ali elektronske lastnosti ali drugi fiz-kem parametri molekule/skupine
- Zamenjave funkcionalnih skupin , ki ne ustrezano klasični definiciji; ni nujno, da imajo isto št.atomov
- Večina spada med neklasične BI

Neklasični bioizosteri funkcionalnih skupin

Karboksilna skupina

- $pK_a \sim 4-5$
- H-vezi

Direktni bioizosteri

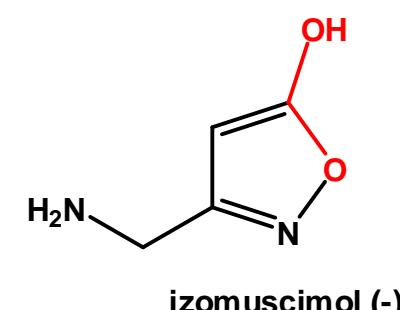
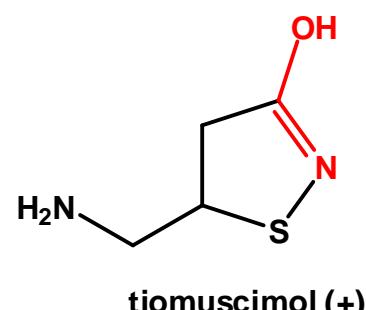
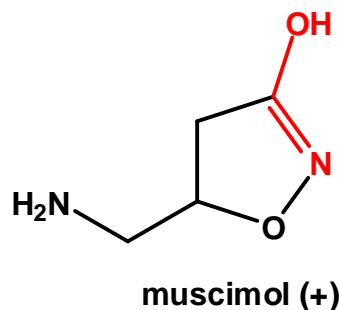
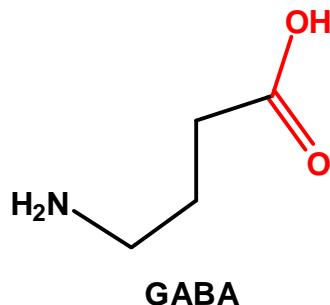
- **Hidroksamska skupina, (acil-)sulfonamidi**
- **Ibuprofen-ibuproksan**

Neklasični bioizosteri funkcionalnih skupin

Karboksilna skupina

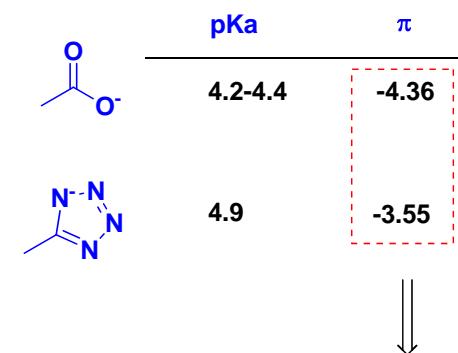
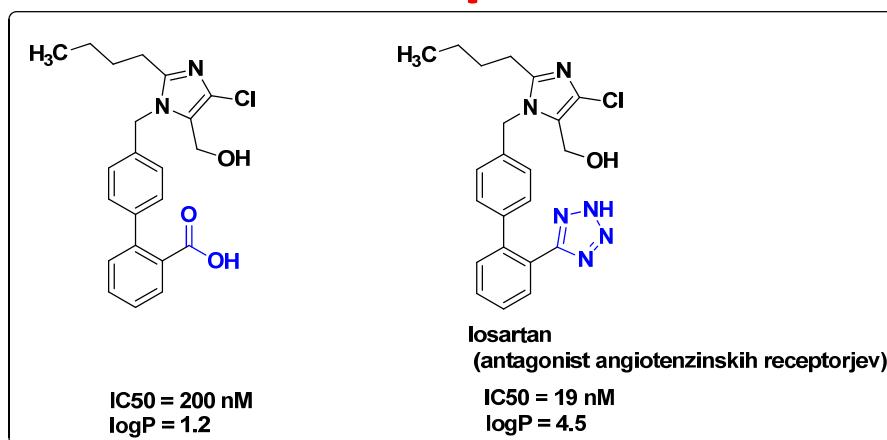
Planarni obročni bioizosteri

- GABA agonisti

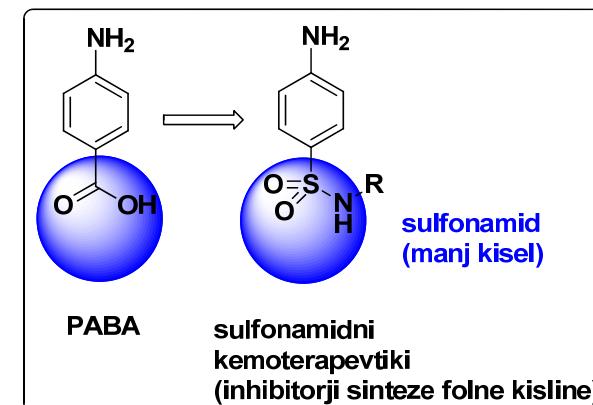
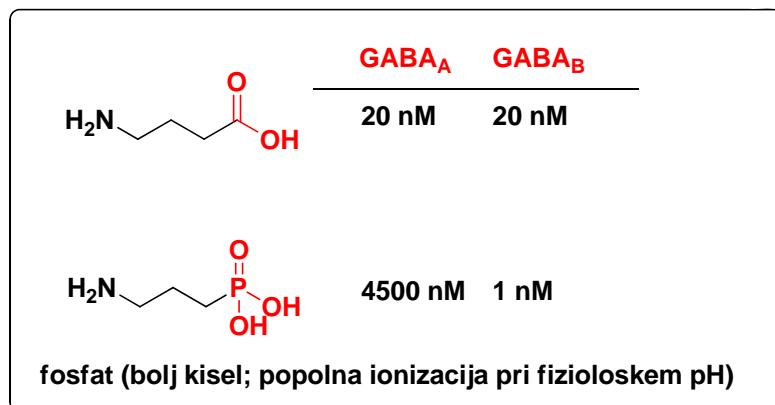


Neklasični bioizosteri funkcionalnih skupin

Karboksilna skupina



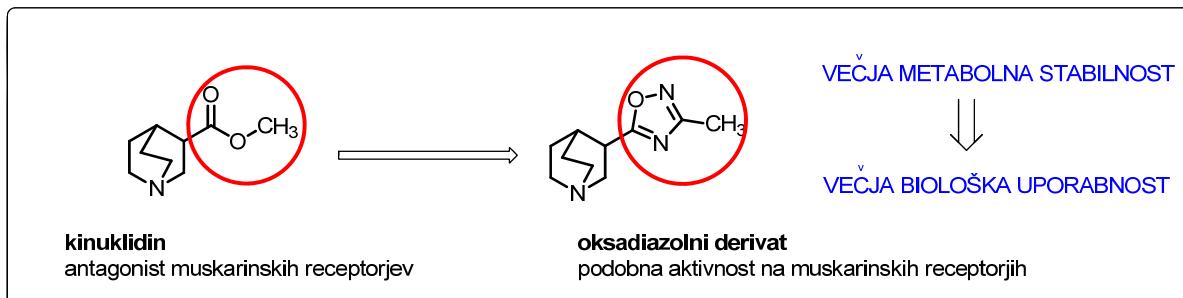
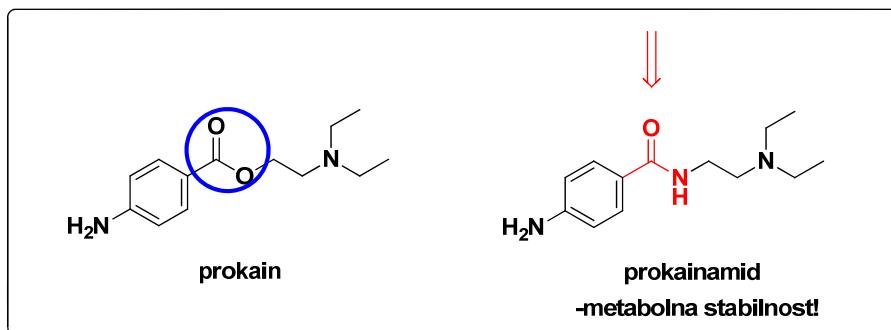
tetrazolni anion skoraj 10x bolj lipofilen!!!



Neklasični bioizosteri funkcionalnih skupin

Estri

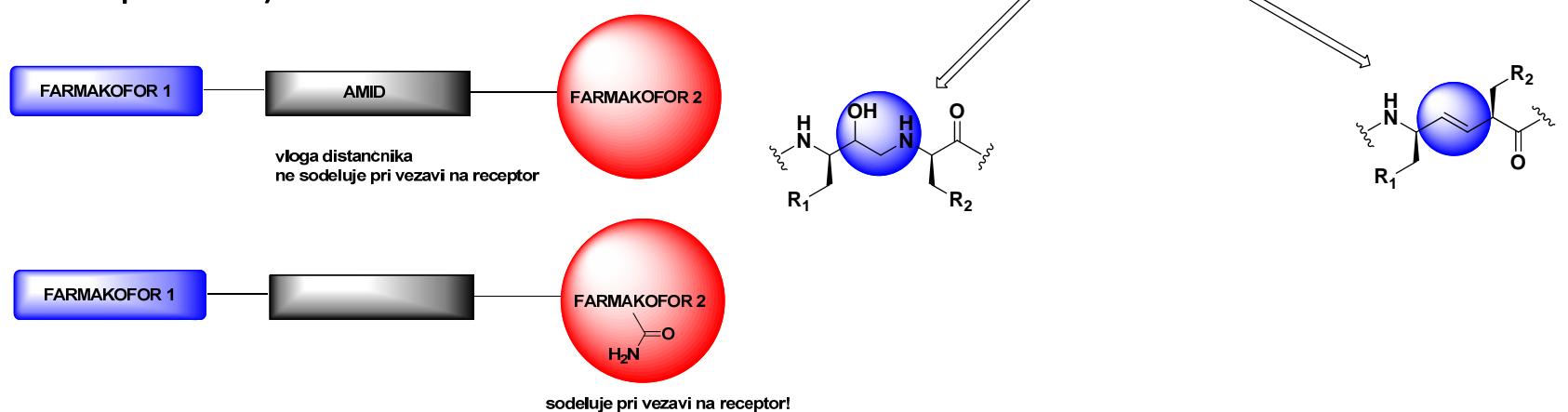
- Labilni *in vivo*; metabolično stabilni analogi
- Prokain - prokainamid



Neklasični bioizosteri funkcionalnih skupin

(Karboks)amidi

- Cela vrsta bioizosterov – naslednje predavanje
- Razgradnja amidov v organizmu
- Plazemske amidaze! → labilnost amidov → slaba BU
- Peptidne ZU (problem peroralne uporabe)

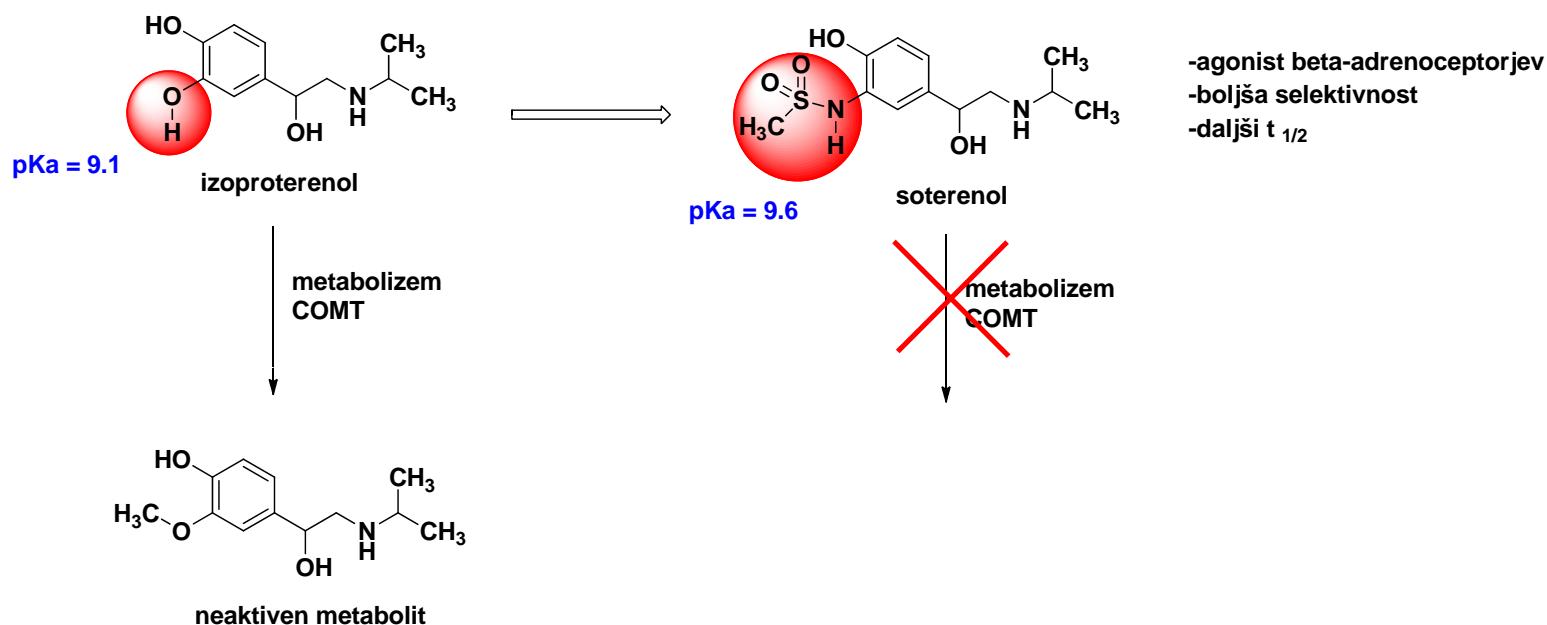


Neklasični bioizosteri funkcijskih skupin

Fenoli

- Izoproterenol – soterenol

Adrenergični sistem:

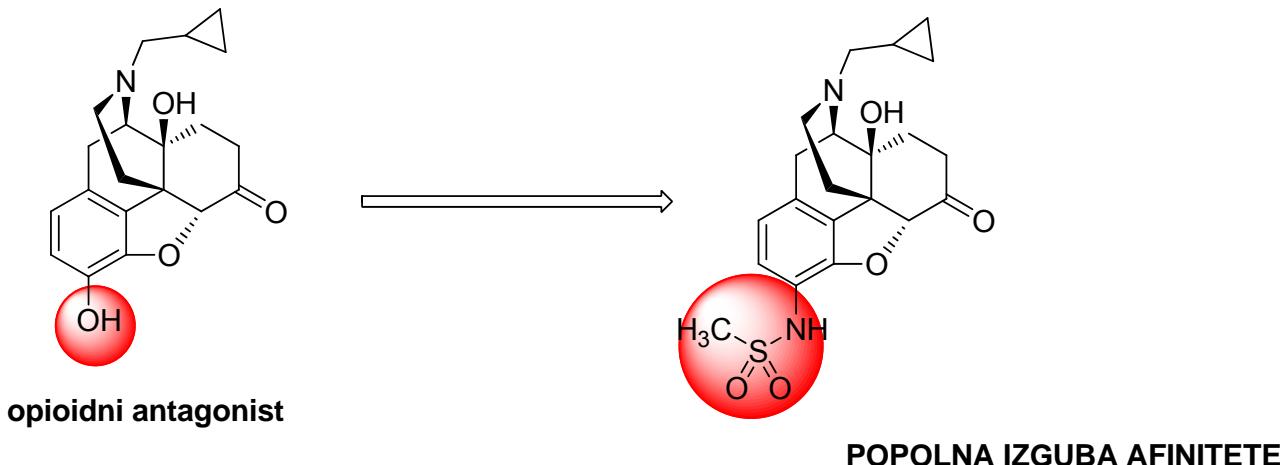


Neklasični bioizosteri funkcijskih skupin

Fenoli

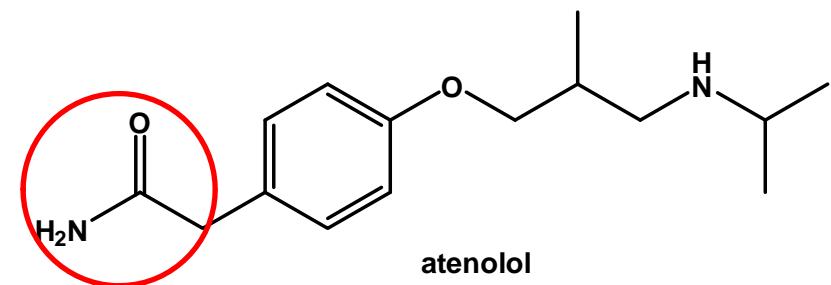
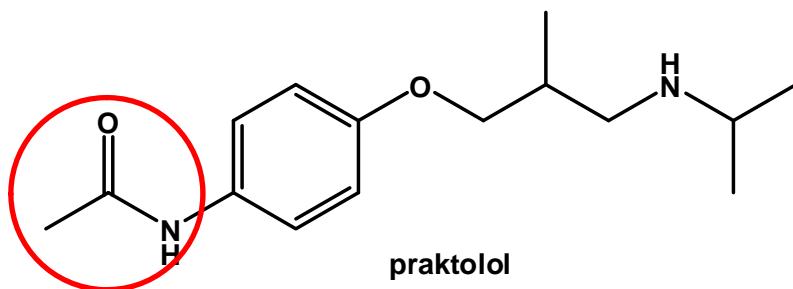
- -OH \rightarrow $-\text{SO}_2\text{NH}$; ni zagotovilo za uspeh!

Opioidni sistem:



Retro-bioizosteri

- β -blokatorji praktolol – atenolol



Ciklični – neciklični analogi

Zamenjava cikličnega dela molekule z necikličnim – posnemanje iste prostorske usmeritve/razporeditve

- Primer estradiol – dietilstilbestrol, heksestrol

Ciklični – neciklični analogi

Primer - propranolol (izumitelj sir James W. Black – Nobelova nagrada 1988)

“Bioizosteri” kitajskega porekla

- Izosteri – podobna analitika
- <http://www.sciencedaily.com/releases/2008/04/080423171529.htm>

The NEW ENGLAND JOURNAL of MEDICINE

ORIGINAL ARTICLE

Contaminated Heparin Associated with Adverse Clinical Events and Activation of the Contact System

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ABSTRACT

BACKGROUND

There is an urgent need to determine whether oversulfated chondroitin sulfate (OSCS), a compound contaminating heparin supplies worldwide, is the cause of the severe anaphylactoid reactions that have occurred after intravenous heparin administration in the United States and Germany.

The figure displays two chemical structures of glycosaminoglycan (GAG) chains, specifically heparin and oversulfated chondroitin sulfate (OSCS). Both structures are shown as repeating units enclosed in brackets with a subscript 'n'. Each unit consists of two pyranose rings linked by a 1,4-glycosidic bond. In the top structure (heparin), the 1-position of the left ring is substituted with an acetyl group (CH₃CO-), the 2-position with a glucuronic acid group (-COOH), the 3-position with an O-sulfate group (-OSO₃H), and the 4-position with an O-sulfate group (-OSO₃H). The 6-position of the left ring is free. The 1-position of the right ring is substituted with a glucosamine group (-NH-CH₂-COOH), the 2-position with a CH₂OR₁ group, and the 3-position with an O-sulfate group (-OSO₃H). In the bottom structure (OSCS), the 2-position of the left ring is substituted with a hydroxyl group (-OH) instead of a glucuronic acid group. The 3-position of the left ring is substituted with an OSO₃H group instead of a CH₂OR₁ group. The 6-position of the left ring is free.

Sklepi

- Razmeroma stroga pravila pri klasičnih bioizosterih
- Pri neklasičnih bioizosterih obratno
- Velika vloga bioizosterov pri razvoju ZU (jakost, ADME, str.učinki); spremembe funkc.skupin (ne celih molekul)
- Uporaba bioizosternih zamenjav odvisna od dane farmakološke skupine!
- Ni preprostih pravil! (izkušnje, intuicija)