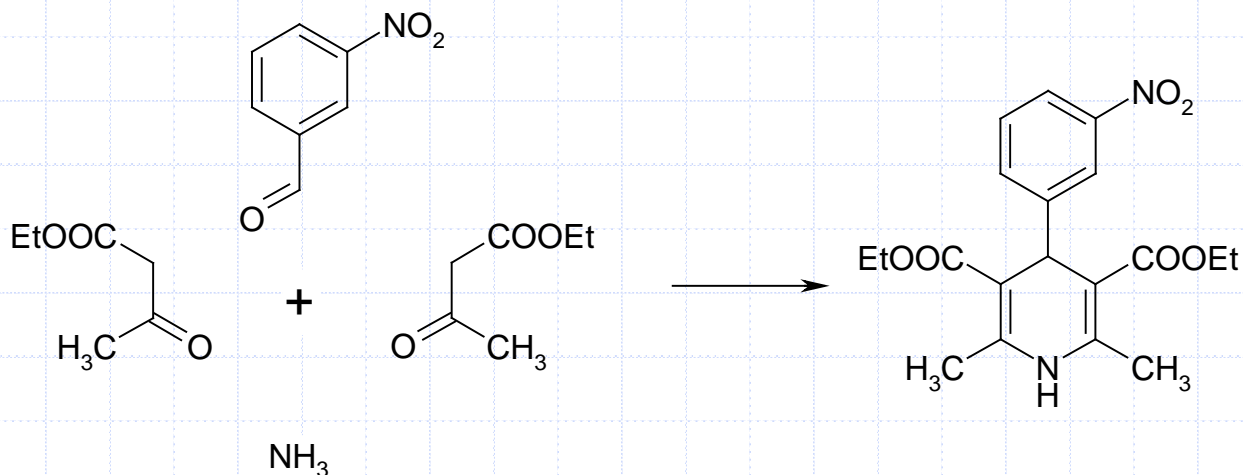




Katedra za farmacevtsko kemijo

Ca-antagonist

Pregled sinteznega postopka



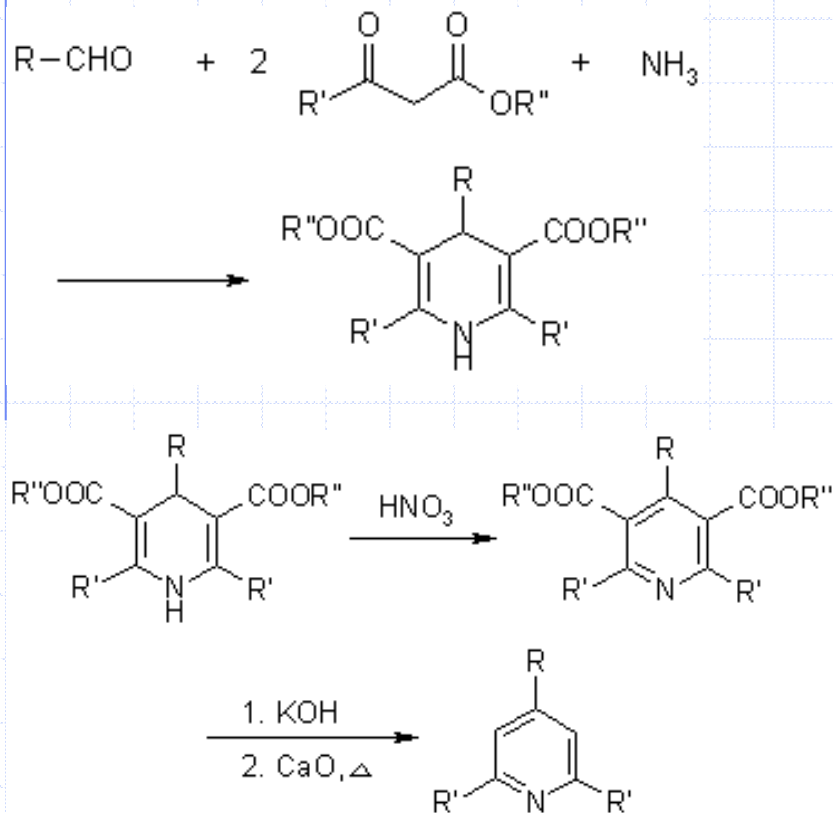
Multikomponentna reakcija:

Aldehid + amin + 2 ekv β -keto-ester

Hantzsch, 1881

Mehanizem Hantzscheve sinteze

Hantzsch Dihydropyridine (Pyridine) Synthesis

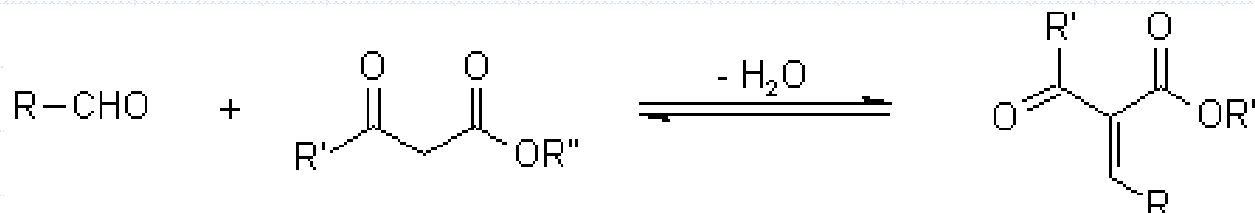


This reaction allows the preparation of dihydropyridine derivatives by condensation of an aldehyde with two equivalents of a β -ketoester in the presence of ammonia. Subsequent oxidation (or dehydrogenation) gives pyridine-3,5-dicarboxylates, which may also be decarboxylated to yield the corresponding pyridines.

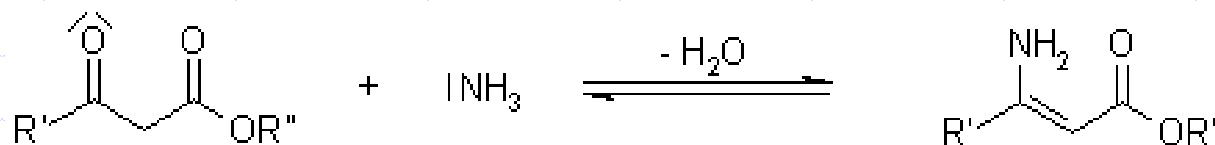
<http://www.organic-chemistry.org/namedreactions/hantzsch-dihydropyridine-synthesis.shtm>

Mehanizem Hantzscheve sinteze

1. Aldehyd + 1 ekv β-keto estra: Knoevenaglova kondenzacija

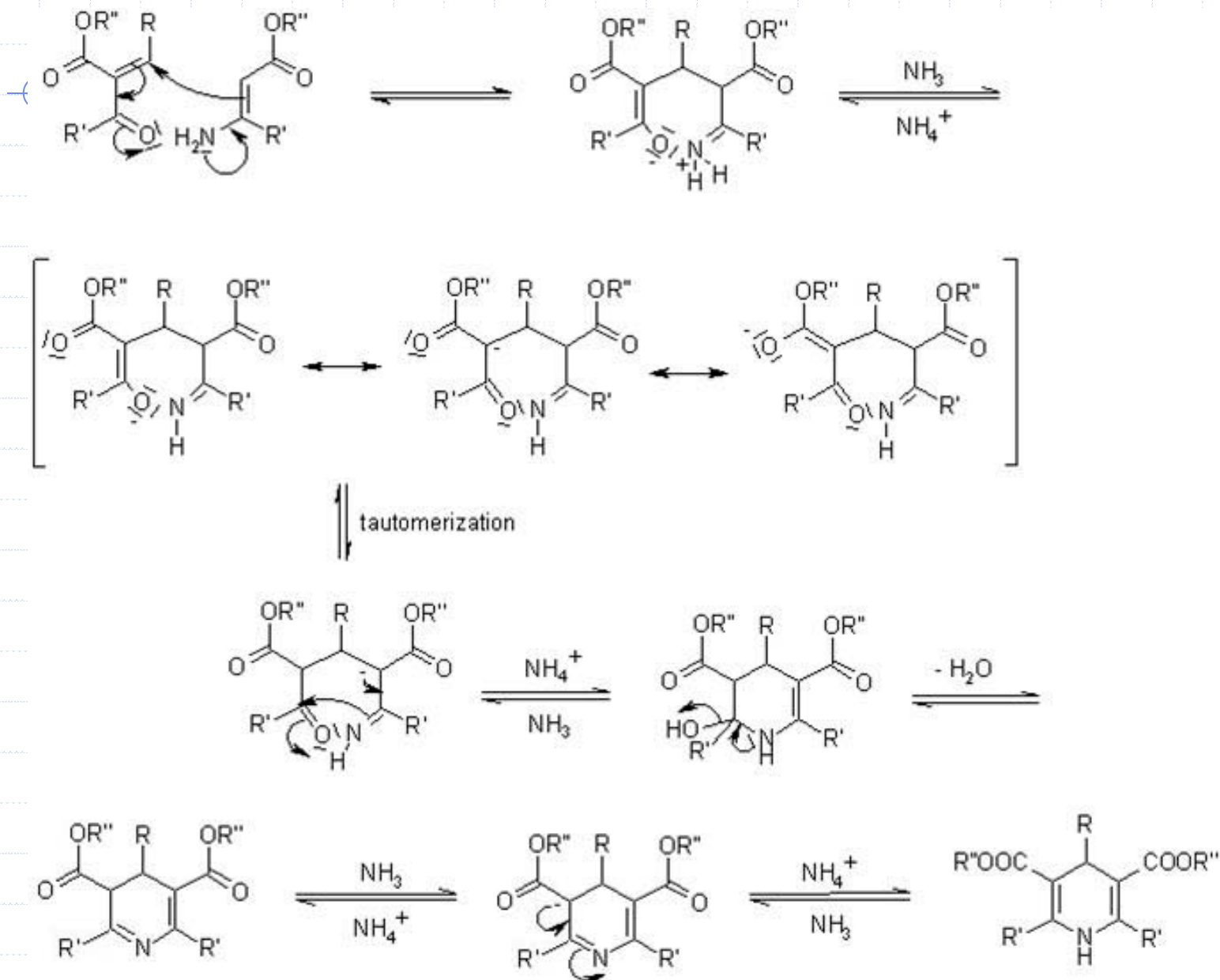


2. Amin + 1 ekv β-keto estra: adicija s sledečo eliminacijo: tvorba imina (tavt. enamina)



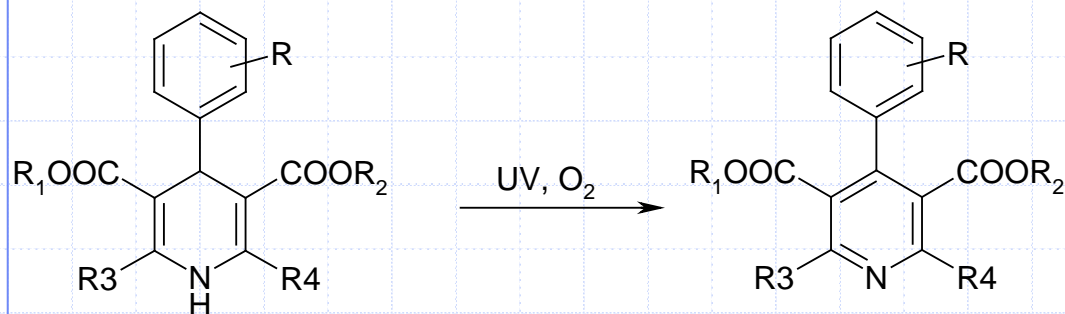
<http://www.organic-chemistry.org/namedreactions/hantzsch-dihydropyridine-synthesis.shtm>

3. Michaelova adicija (1,4-adicija) enamina na α,β -nenasičen keton (prebiten NH_3 kot katalizator)



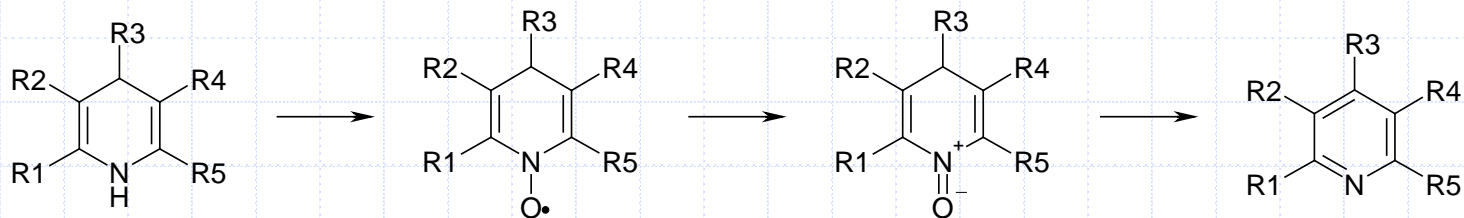
Lastnosti 1,4-dihidropiridinov

Občutljivi na oksidante: oksidacija do piridinov



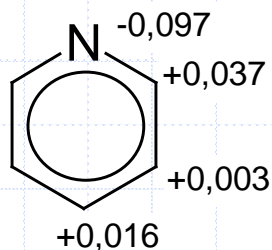
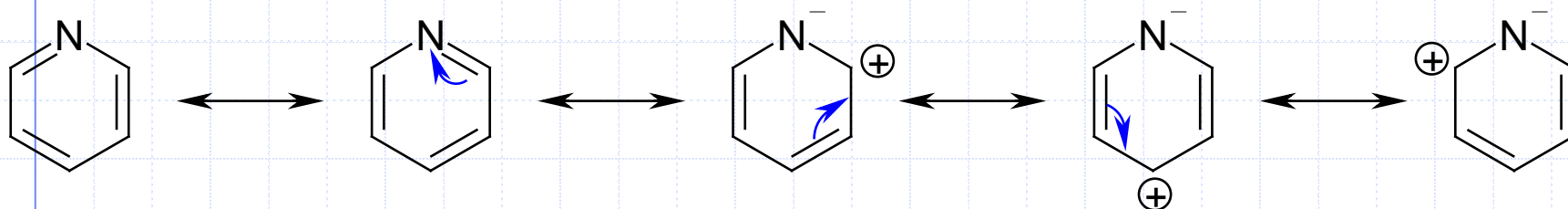
Učinkovine zaščitene pred svetlobo!

Mehanizem: preko nitroksidnega radikala



Reaktivnost piridinov

Manjša elektronska gostota obročnih C zaradi prisotnosti bolj elektronprivlačnega N atoma



⇒ manjša reaktivnost za elektrofilne aromatske substitucije

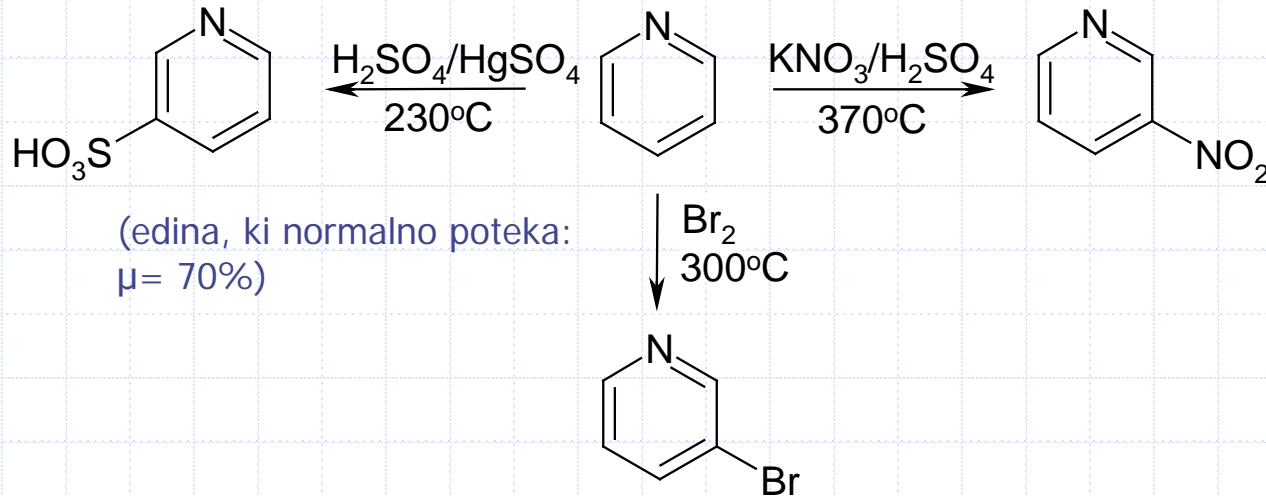
⇒ SE_{Ar} potekajo na m-mesto

⇒ bolj občutljiv za nukleofile

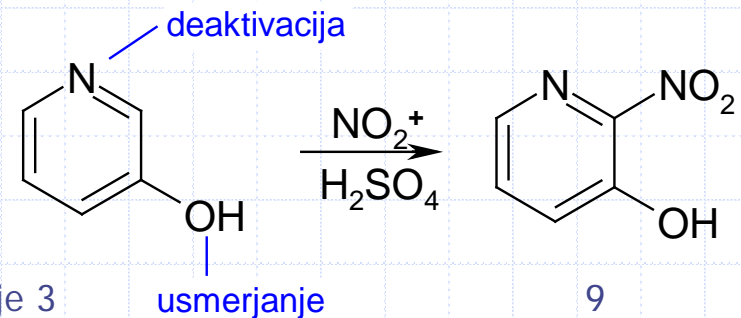
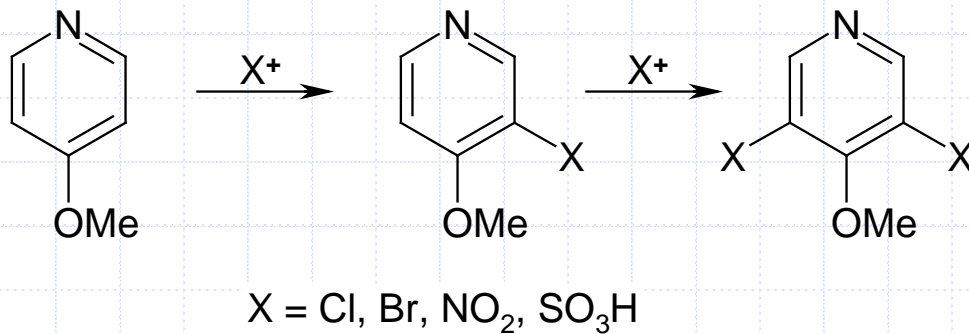
⇒ SN_{Ar} potekajo na o- in p- mesto

Reaktivnost piridinov

SE_{Ar} : zgolj pod zelo ostrimi reakcijskimi pogoji



Skupine, ki aktivirajo za SE_{Ar} :

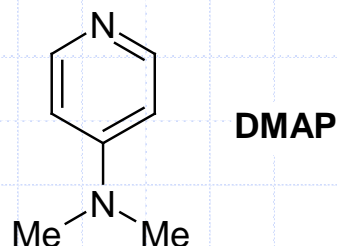
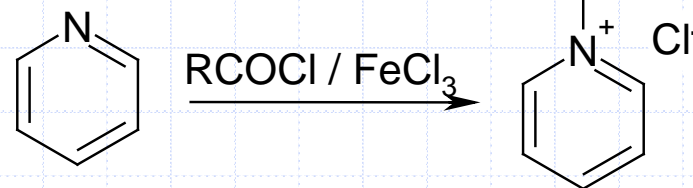
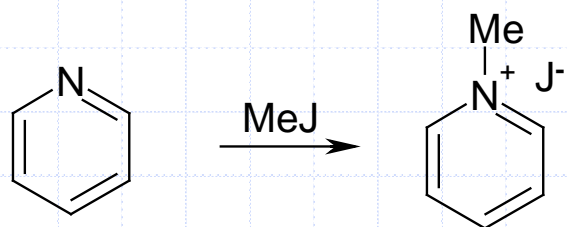


Reaktivnost piridinov

Friedel-Crafts-ove reakcije ne potekajo!

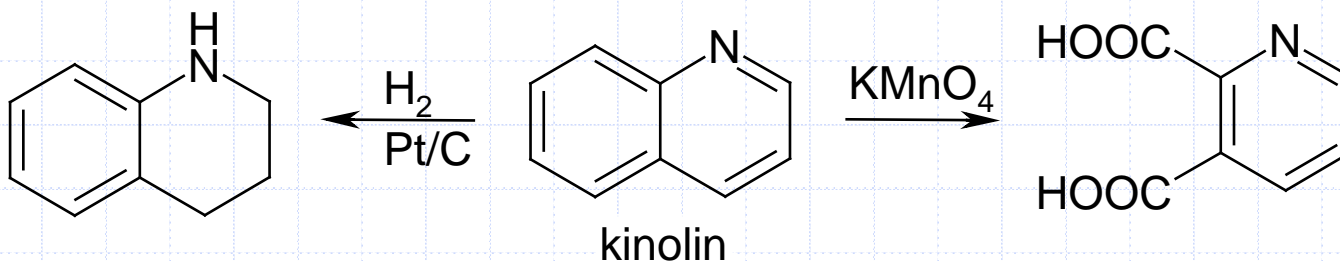
Alkiliranje in aciliranje poteče na obročni N-atom

→ piridinijeve soli



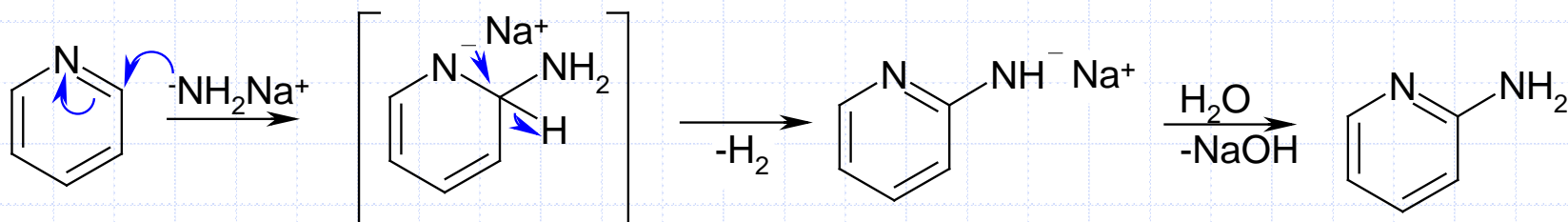
Reaktivnost piridinov

Piridin je bolj obstojen napram oksidacijam kot benzen, lažje pa ga reduciramo (velja za vse heteroaromate):

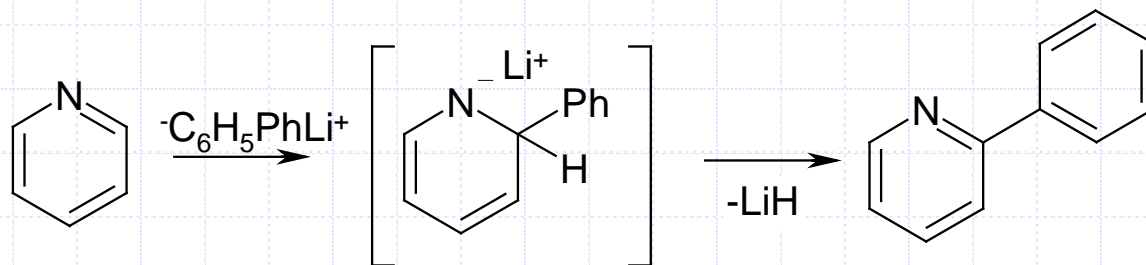


Reaktivnost piridinov: S_NAr

Čičibabin-ova reakcija: aminiranje na o-mestu z natrijevim amidom

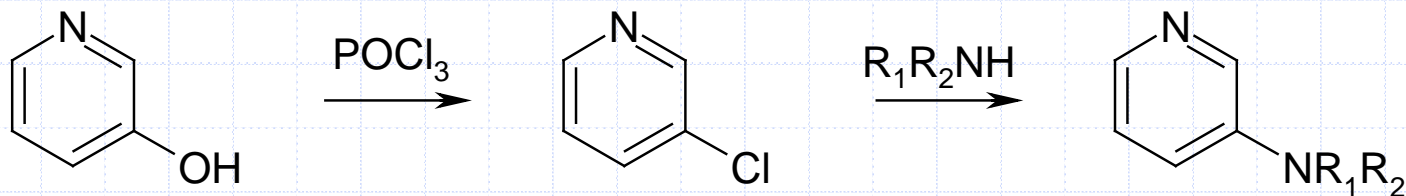
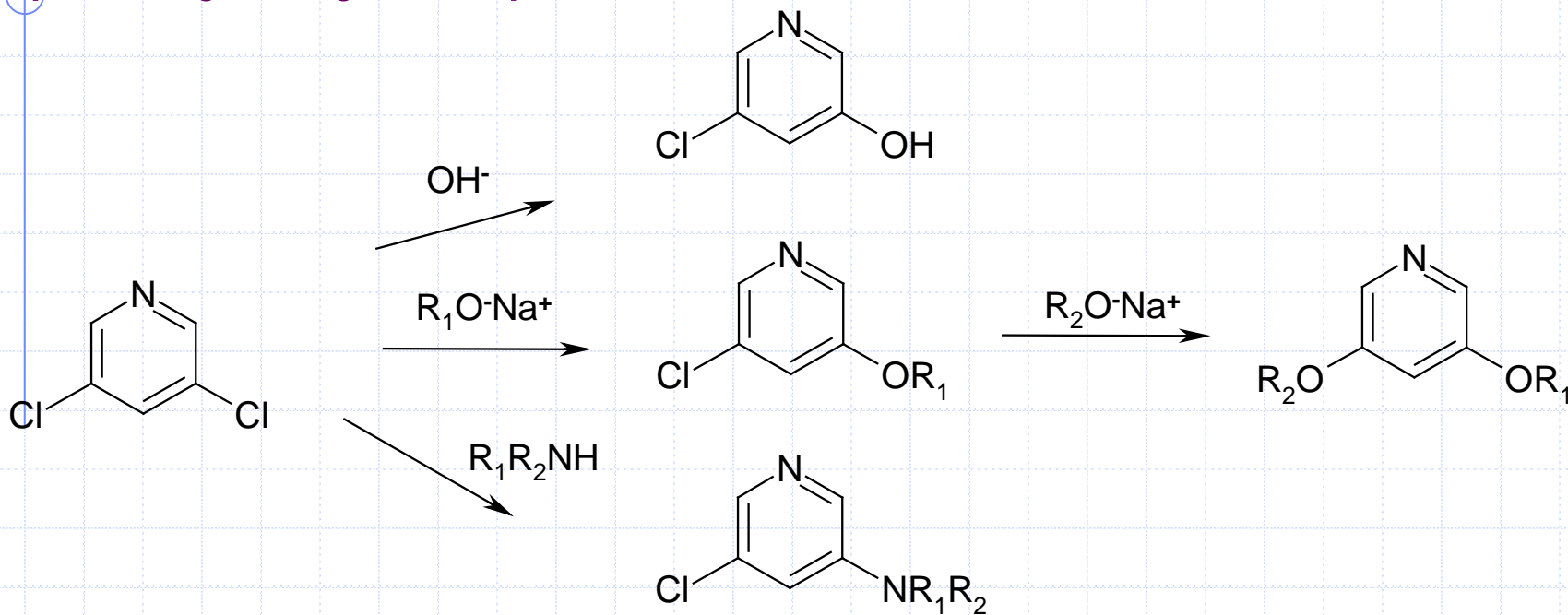


Analogno: reakcija s fenil litijem



Nukleofilne aromatske substitucije

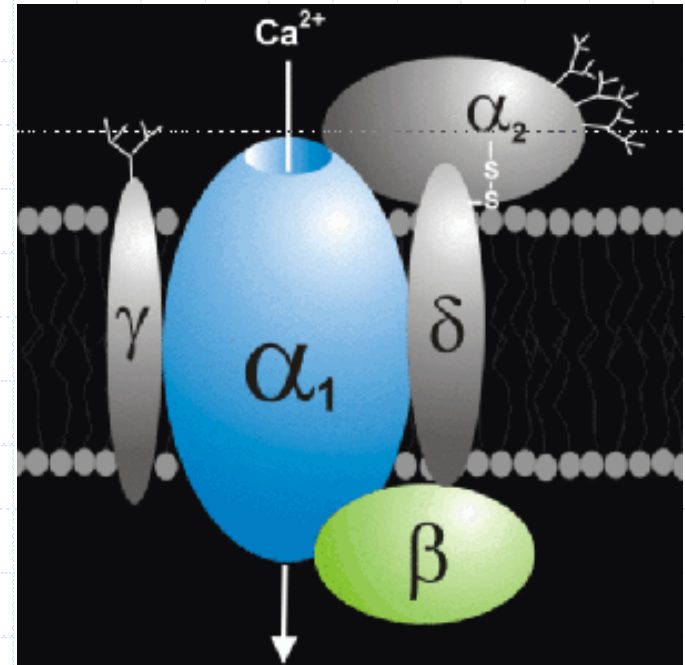
potekajo lažje kot pri benzenu



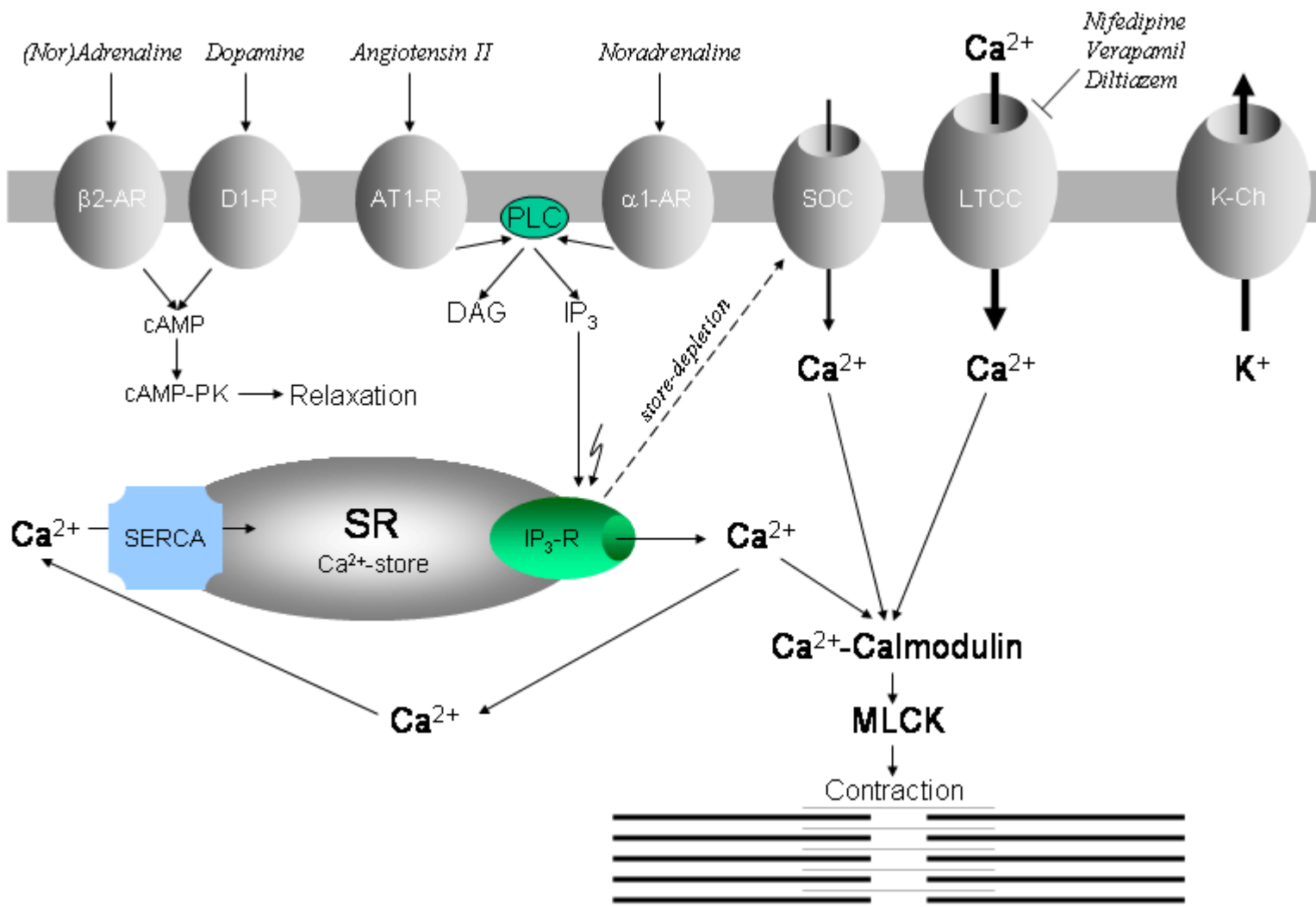
Ca antagonisti

◆ SKRIPTA

- mehanizem delovanja
- skupine kalcijevih antagonistov
- metabolizem
- odnos med strukturo in delovanjem pri dihidropiridinski skupini



Ca antagonisti



Ca antagonisti

- ◆ *Simplified view of the pharmacological action of L-type Ca²⁺ channel blockers in arterial smooth muscle: In contrast to cardiomyocytes action potentials are not carried by fast sodium channels in smooth muscle and depolarizations are more long lasting. Contraction requires the binding of Ca²⁺ to calmodulin, which then activates myosin light chain kinase (MLCK). MLCK phosphorylates the light chain of myosin which turns on contraction. The Ca²⁺ for activation of this pathway can enter through L-type Ca²⁺ channels in response to depolarization. Ca²⁺ channel blockers inhibit this pathway through concentration-dependent block of Ca²⁺ entry. Alternatively, Ca²⁺ can be released from intracellular stores after activation of membrane receptors (e.g. of angiotensin II AT1 or α_1 -adrenergic receptors) coupled to IP₃ production. IP₃ opens IP₃ receptor channels, RyR related Ca²⁺ release channels in the SR. This process does not involve L-type Ca²⁺ channels and is not inhibited by Ca²⁺ channel blockers. Store-depletion also triggers the activation of "store-operated channels" (SOC) in the plasma membrane which are also not sensitive to Ca²⁺ channel blockers. Receptor-mediated activation of cAMP-dependent protein kinase (cAMP-PK) results in muscle relaxation through different mechanisms. D1-R, dopamine1 receptor; AR, adrenergic receptor; PLC, phospholipase C.*

Sintezne naloge

◆ Sinteza NIKARDIPINA

3-{2-[benzil(metil)amino]etil}-5-metil-2,6-dimetil-4-(3-nitrofenil)-1,4-dihidro-3,5-piridindikarboksilat

Iz metilacetoacetata, amoniaka in 2-[benzil(metil)amino]etil-3-oksobutanoata

◆ Sinteza NIFEDIPINA

dimetil 2,6-dimetil-4-(2-nitrofenil)-1,4-dihidro-3,5-piridindikarboksilat

Iz metil 3-oksobutanoata, amoniaka in 2-nitrobenzaldehyda

Sintezne naloge

◆ Sinteza VERAPAMILA

4-[(3,4-dimetoksifenetil)metilamino]-1-(3,4-dimetoksifenil)-1-izopropil-butilcianid

Iz 1-(3,4-dimetoksifenil)benzilcianida, 1-bromo-3-kloropropana in N-[2-(3,4-dimetoksifenil)etil]-N-metilamina

Naloga za točko