



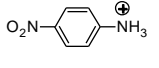
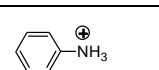
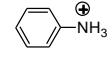
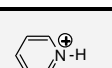
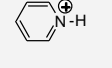
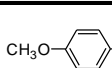
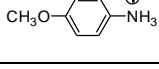
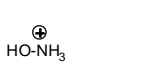
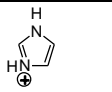
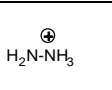
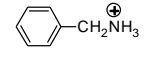
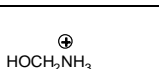


OH KISLINE		
-12.	FSO ₃ H	FSO ₃ [⊖]
-12	RNO ₂ [⊕] H	RNO ₂ H
-10	HClO ₄	ClO ₄ [⊖]
-9	H ₂ SO ₄	HSO ₄ [⊖]
-8		
-7		
-6.5	ArSO ₃ H	ArSO ₃ [⊖]
-6.5		
-6.4		Ar-O _H
-6	CH ₃ SO ₃ H	CH ₃ SO ₃ [⊖]
-6		
-6		Ar-O _R
-3.5		R-O _R
-2.4	CH ₃ CH ₂ OH ₂ [⊕]	CH ₃ CH ₂ OH
-1.7	H ₃ O [⊕]	H ₂ O
-1.5		
-1.5	(CH ₃) ₂ S=OH [⊕]	(CH ₃) ₂ S=O
-1.4	HNO ₃	NO ₃ [⊖]
-0.5		
0.5	F ₃ C-COOH	F ₃ C-COO [⊖]
0.7		

1.5		
1.7		
1.8		
2.0	HSO ₄ [⊖]	SO ₄ ^{2⊖}
2.2	H ₃ PO ₄	H ₂ PO ₄ [⊖]
2.4		
2.5	N≡C-CH ₂ -OH	N≡C-CH ₂ -O [⊖]
2.5		
2.9	Cl-CH ₂ -COOH	Cl-CH ₂ -COO [⊖]
3.1	F ₃ CH ₂ -COOH	F ₃ CH ₂ -COO [⊖]
3.3	HO-N=O	⊖O-N=O
3.4		
3.6		
3.6	H ₃ COH ₂ C-COOH	H ₃ COH ₂ C-COO [⊖]
4.2		
4.5		
4.6	(CH ₃) ₃ N [⊕] -OH	(CH ₃) ₃ N [⊕] -O [⊖]
4.8		
6.4	H ₂ CO ₃	HCO ₃ [⊖]
7.2	H ₂ PO ₄ [⊖]	HPO ₄ ^{2⊖}
7.2		
10.2		

10.3	HCO ₃ [⊖]	CO ₃ ^{2⊖}
11.6	HO-OH	⊖O-OH
12.2	(CH ₃) ₂ C=N-OH	(CH ₃) ₂ C=N-O [⊖]
12.4	HPO ₄ ^{2⊖}	PO ₄ ^{3⊖}
12.4	CF ₃ CH ₂ OH	CF ₃ CH ₂ O [⊖]
13.3	HOCH ₂ OH	HOCH ₂ O [⊖]
14.2	HOCH ₂ CH ₂ O [⊖]	HOCH ₂ CH ₂ O [⊖]
15.5	CH ₃ OH	CH ₃ O [⊖]
15.7	H ₂ O	⊖OH
16	CH ₃ CH ₂ OH	CH ₃ CH ₂ O [⊖]
18	(CH ₃) ₂ CHOH	(CH ₃) ₂ CHO [⊖]
19	(CH ₃) ₃ COH	(CH ₃) ₃ CO [⊖]
CH KISLINE		
-5	HC(CN) ₃	⊖C(CN) ₃
3.6	H ₂ C(NO ₂)	⊖HC(NO ₂) ₂
5		
9		
9	N≡C-CH ₂ -C(=O)CH ₃	N≡C-⊖CH-C(=O)CH ₃
9.2	N≡CH	N≡C⊖
10		
10.2	H ₃ C-NO ₂	H ₂ C-NO ₂ [⊖]
10.7		

12.5	$\text{H}_2\text{C}(\text{SO}_2\text{CH}_3)_2$	$\ominus\text{HC}(\text{SO}_2\text{CH}_3)_2$
13	$\text{H}_2\text{C}(\text{-C-OEt})_2$	$\ominus\text{HC}(\text{-C-OEt})_2$
13.5	$\text{H}_3\text{C}-\text{C}(\text{O})-\text{O}-\text{C}(\text{O})-\text{CH}_3$	$\text{H}_2\text{C}-\text{C}(\text{O})-\text{O}-\text{C}(\text{O})-\text{CH}_3$
13.6	HCCl_3	$\ominus\text{CCl}_3$
~14	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{SR}$	$\text{H}_2\text{C}-\text{C}(=\text{O})-\text{SR}$
15.9	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{CH}_2\text{Ph}$	$\text{H}_2\text{C}-\text{C}(=\text{O})-\text{CH}_2\text{Ph}$
16		
16	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{CH}_2\text{Cl}$	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{CHCl}$
~16	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{Cl}$	$\text{H}_2\text{C}-\text{C}(=\text{O})-\text{Cl}$
16.7	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{H}$	$\text{H}_2\text{C}-\text{C}(=\text{O})-\text{H}$
19.2	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{CH}_3$	$\text{H}_2\text{C}-\text{C}(=\text{O})-\text{CH}_3$
23	$\text{H}_3\text{C}-\text{SO}_2\text{CH}_3$	$\text{H}_2\text{C}-\text{SO}_2\text{CH}_3$
24	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{OR}$	$\text{H}_2\text{C}-\text{C}(=\text{O})-\text{OR}$
25	$\text{H}_3\text{C}-\text{C}\equiv\text{N}$	$\text{H}_2\text{C}-\text{C}\equiv\text{N}$
25	$\text{HC}\equiv\text{CH}$	$\ominus\text{C}\equiv\text{CH}$
~28	$\text{H}_3\text{C}-\text{C}(=\text{O})-\text{NR}_2$	$\text{H}_2\text{C}-\text{C}(=\text{O})-\text{NR}_2$
31.1		
31.5	Ph_3CH	$\text{Ph}_3\text{C}^\ominus$
33.5	Ph_2CH_2	$\text{Ph}_2\text{CH}^\ominus$
35	$\text{H}_3\text{C}-\overset{\text{O}}{\parallel}{\text{S}}-\text{CH}_3$	$\text{H}_2\text{C}-\overset{\text{O}}{\parallel}{\text{S}}-\text{CH}_3$
35	$\text{CH}_3-\text{P}^+\text{Ph}_3$	$\text{CH}_2-\text{P}^+\text{Ph}_3$
41	$\text{H}_3\text{C}-\text{Ph}$	$\text{H}_2\text{C}-\text{Ph}$

43	$\text{H}_3\text{C}-\text{HC}=\text{CH}_2$	$\text{H}_2\text{C}-\text{HC}=\text{CH}_2$
43	HPh	$\ominus\text{Ph}$
44	$\text{H}_2\text{C}=\text{CH}_2$	$\text{HC}=\text{CH}_2$
48	CH_4	$\ominus\text{CH}_3$
50	$\text{H}_3\text{C}-\text{CH}_3$	$\text{H}_2\text{C}-\text{CH}_3$
51	$\text{H}_2\text{C}(\text{CH}_3)_2$	$\text{HC}(\text{CH}_3)_2$
>52	$\text{HC}(\text{CH}_3)_3$	$\ominus\text{C}(\text{CH}_3)_3$
NH KISLINE		
-10	$\text{RC}\equiv\text{NH}^\oplus$	$\text{RC}\equiv\text{N}$
-5	$\text{Ar}_3\text{NH}^\oplus$	Ar_3N
1	$\text{Ar}_2\text{NH}_2^\oplus$	Ar_2NH
1.0		
4.6		
5.2		
5.4		
5.8	$\text{HO}-\text{NH}_3^\oplus$	$\text{HO}-\text{NH}_2$
7.0		
7.9	$\text{H}_2\text{N}-\text{NH}_3^\oplus$	$\text{H}_2\text{N}-\text{NH}_2$
8.5	PhSO_2NH_2	$\text{PhSO}_2\text{NH}^\ominus$
9.2	NH_4^\oplus	NH_3
9.3		
5.8	$\text{HOCH}_2\text{NH}_3^\oplus$	HOCH_2NH_2

9.8		
10.6	EtNH_3^\oplus	EtNH_2
10.7	$\text{Et}_3\text{NH}^\oplus$	Et_3N
11	$\text{Et}_2\text{NH}_2^\oplus$	Et_2NH
13.6	$(\text{NH}_2)_2\text{C}=\text{NH}_2^\oplus$	$(\text{NH}_2)_2\text{C}=\text{NH}$
17		
25.8	$((\text{CH}_3)_3\text{Si})_2\text{NH}$	$((\text{CH}_3)_3\text{Si})_2\text{N}^\ominus$
27	PhNH_2	PhNH^\ominus
35	NH_3	NH_2^\ominus
36	Et_2NH	$\text{Et}_2\text{N}^\ominus$
OSTALE KISLINE		
-10	HI	I^\ominus
-9	HBr	Br^\ominus
-7	HCl	Cl^\ominus
-7	RSH_2^\oplus	RSH
-5.3	$\text{R}_2\text{SH}^\oplus$	R_2S
3.2	HF	F^\ominus
3.3	$\text{H}_3\text{C}-\text{SH}$	$\text{H}_3\text{C}-\text{S}^\ominus$
3.9	H_2Se	HSe^\ominus
7.0	H_2S	HS^\ominus
10.6	EtSH	EtS^\ominus
35	H_2	H^\ominus

Average Single Bond Energies In kcal/mol (kJ/mol)

	I	Br	Cl	S	P	Si	F	O	N	C	H		
H	71 (297)	87 (364)	103 (431)	83 (347)	77 (322)	76 (318)	135 (565)	111 (464)	93 (389)	99 (414)	104 (435)		
C	51 (213)	68 (285)	81 (339)	65 (272)	63 (264)	72 (301)	116 (485)	86 (360)	73 (305)	83 (347)			
N			46 (192)				65 (272)	53 (222)	39 (163)				
O	48 (201)	48 (201)	52 (218)		141 (590)	108 (452)	45 (188)	47 (197)					
F	58 (243)	60 (251)	61 (255)	68 (285)	117 (490)	135 (565)	37 (155)						
Si	56 (234)	74 (310)	91 (381)			53 (222)							
P	44 (184)	63 (264)	78 (326)		48 (201)				C=C	146 (611)		C≡C	200 (837)
S		52 (218)	61 (255)	60 (251)					C=N	147 (615)		C≡N	213 (891)
Cl	50 (209)	52 (218)	58 (243)						C=O	177 (741)			
Br	42 (176)	46 (192)							C=S	128 (536)			
I	36 (151)								N=N	100 (418)		N≡N	226 (946)
									N=O	145 (607)			
									O=O	119 (498)			

Table 2.1 ΔG° and K_{eq} Values for 25°C

ΔG° kcal/mol	K_{eq}	Reactant	Product	ΔG° (kJ/mol)
+5.46	0.0001	99.99	0.01	+22.84
+4.09	0.001	99.9	0.1	+17.11
+2.73	0.01	99	1	+11.42
+1.37	0.1	91	9	+5.73
+1.0	0.18	85	15	+4.18
+0.5	0.43	70	30	+2.09
0	1	50	50	0
-0.5	2.33	30	70	-2.09
-1.0	5.41	15	85	-4.18
-1.37	10	9	91	-5.73
-2.73	100	1	99	-11.42
-4.09	1,000	0.1	99.9	-17.11
-5.46	10,000	0.01	99.99	-22.84
-9.56	10^7	Essentially complete		-40.00

Substituent X	pK_a <i>p</i> -X-benzojska k.	pK_a <i>m</i> -X-benzojska k.	σ_p	σ_m
NH ₂	4.82	4.20	-0.62	0.00
OCH ₃	4.49	4.09	-0.29	0.11
CH ₃	4.37	4.26	-0.17	-0.06
H	4.20	4.20	0.00	0.00
F	4.15	3.86	0.05	0.34
I	3.97	3.85	0.23	0.35
Cl	3.98	3.83	0.22	0.37
Br	3.97	3.80	0.23	0.40
CO ₂ CH ₃	3.75	3.87	0.45	0.33
COCH ₃	3.71	3.83	0.49	0.37
CN	3.53	3.58	0.67	0.62
NO ₂	3.43	3.47	0.77	0.73

Solvent	$f_p^{(c)}$ (°C)	$bp^{(d)}$ (°C)	$\epsilon^{(e)}$ at 25 °C	$\mu \cdot 10^{30 \text{ (f)}}$ (C m)	$n_D^{20 \text{ (g)}}$	$E_T(30)^{(h)}$ (kcal/mole)
1. Water	0.0	100.0	78.39	6.07	1.3330	63.1
2. Formamide	2.5	210.5	111.0 (20°)	11.24	1.4475	56.6
3. 1,2-Ethandiol	- 13	197.3	37.7	7.61	1.4318	56.3
4. Methanol	- 97.7	64.7	32.70	5.67	1.3284	55.5
5. <i>N</i> -Methylformamide	- 3.8	180-185	182.4	12.88	1.4319	54.1
6. Diethylene glycol	- 6.5	244.8	31.69 (20°)	7.71	1.4475	53.8
7. Triethylene glycol	- 4.3	288.0	23.69 (20°)	9.97	1.4561	53.5
8. 2-Methoxyethanol	- 85.1	124.6	16.93	6.81	1.4021	52.3
9. <i>N</i> -Methylacetamide	30.6	206.7	191.3 (32°)	14.65	1.4286 (28°)	52.0
10. Ethanol	-114.1	78.3	24.55	5.77	1.3614	51.9
11. 2-Aminoethanol	10.5	171.0	37.72	7.57	1.4539	51.8
12. Acetic acid	16.7	117.9	6.15 (20°)	5.60	1.3719	51.2 ¹⁾
13. Benzyl alcohol	- 15.3	205.5	13.1 (20°)	5.54	1.5404	50.8
14. 1-Propanol	-126.2	97.2	20.33	5.54	1.3856	50.7
15. 1-Butanol	- 88.6	117.7	17.51	5.84	1.3993	50.2
16. 2-Methyl-1-propanol, Isobutyl alcohol	-108	107.7	17.93	5.97	1.3959	49.0
17. 2-Propanol	- 88.0	82.3	19.92	5.54	1.3772	48.6
18. 2-Butanol, <i>sec</i> -Butyl alcohol	-114.7	99.6	16.56	5.54	1.3972	47.1
19. 3-Methyl-1-butanol, Isoamyl alcohol	-117.2	130.5	14.7	6.07	1.4071	47.0
20. Cyclohexanol	25.2	161.1	15.0	6.20	1.4648 (25°)	46.9
21. 4-Methyl-1,3-dioxol- 2-one, Propylene carbonate ^(k)	- 48.8	241.7	65.1	16.7	1.4209	46.6
22. 2-Pentanol, <i>sec</i> -Amyl alcohol		119.0	13.82 (22°)	5.54	1.4064	46.5
23. Nitromethane	- 28.6	101.2	35.87 (30°)	11.88	1.3812	46.3
24. Acetonitrile	- 43.8	81.6	37.5 (20°)	11.48	1.3441	46.0
25. 3-Pentanol	- 75	115.3	13.02 (22°)	5.47	1.4103	45.7
26. Dimethylsulfoxide	18.5	189.0	46.68	13.0	1.4783	45.0
27. Aniline	- 5.98	184.4	6.89 (20°)	5.04	1.5863	44.3
28. Tetrahydrothiophene- 1,1-dioxide, Sulfolane	28.5	287.3	43.3 (30°)	16.05	1.4820 (30°)	44.0
29. Acetic anhydride	- 73.1	140.0	20.7 (19°)	9.41	1.3904	43.9 ¹⁾
30. 2-Methyl-2-propanol, <i>tert</i> -Butyl alcohol	25.8	82.4	12.47	5.54	1.3877	43.9
31. <i>N,N</i> -Dimethyl- formamide	- 61	152.3	37.0	12.88	1.4269 (25°)	43.8
32. <i>N,N</i> -Dimethyl- acetamide	- 20	166.1	37.78	12.41	1.4384	43.7
33. Propionitrile	- 92.8	97.4	27.2 (20°)	11.91	1.3658	43.7
34. 1-Methyl-2- pyrrolidinone ^(m)	- 24.4	204.0-204.8	32.0	13.64	1.4700	42.2
35. Acetone	- 94.7	56.3	20.70	9.54	1.3587	42.2
36. Nitrobenzene	5.8	210.8	34.82	13.44	1.5500 (25°)	42.0
37. Benzointrile	- 12.8	191.1	25.20	13.51	1.5282	42.0

Solvent	$f_p^{(c)}$ (°C)	$bp^{(d)}$ (°C)	$\epsilon^{(e)}$ at 25 °C	$\mu \cdot 10^{30 \text{ (f)}}$ (C m)	$n_D^{20 \text{ (g)}}$	$E_T(30)^{(h)}$ (kcal/mole)
38. 1,2-Diaminoethane	11.3	117.3	12.9	6.34	1.4568	42.0
39. 1,2-Dichloroethane	- 35.7	83.5	10.36	6.20	1.4448	41.9
40. 2-Methyl-2-butanol, <i>tert</i> -Amyl alcohol	- 8.8	102.0	5.82	5.7	1.4049	41.9
41. 2-Butanone	- 86.7	79.6	18.51 (20°)	9.21	1.3788	41.3
42. Acetophenone	19.6	202.0	17.39	9.87	1.5342	41.3
43. Dichloromethane	- 95.1	39.8	8.93	5.17	1.4242	41.1
44. 1,1,3,3-Tetramethyl- urea	- 1.2	175.2	23.45 ¹⁾	11.58	1.4493 (25°)	41.0
45. Hexamethylphos- phoric triamide ^(l)	7.2	235	29.6	18.48	1.4584	40.9
46. Cyclohexanone	- 32.1	155.7	18.3 (20°)	10.04	1.4510	40.8
47. Pyridine	- 41.6	115.3	12.4 (21°)	7.91	1.5102	40.2
48. Methyl acetate	- 98.1	56.3	6.68	5.37	1.3614	40.0
49. 4-Methyl-2- pentanone	- 84.0	116.5	13.11 (20°)		1.3957	39.4
50. 1,1-Dichloroethane	- 97.0	57.3	10.0 (18°)	6.61	1.4164	39.4
51. Quinoline	- 14.9	237.1	9.00	7.27	1.6273	39.4
52. 3-Pentanone	- 38.9	102.0	17.00 (20°)	9.41	1.3923	39.3
53. Chloroform	- 63.6	61.2	4.81 (20°)	3.84	1.4429 (25°)	39.1
54. Triethylene glycol dimethyl ether, Triglyme		222	7.5		1.4233	38.9
55. Diethylene glycol dimethyl ether, Diglyme		159.8		6.57	1.4097	38.6
56. 1,2-Dimethoxyethane, Monoglyme	- 58	85 ¹⁾	7.20	5.70	1.3796	38.2
57. 1,2-Dichlorobenzene	- 17.0	180.5	9.93	7.57	1.5515	38.1
58. Ethyl acetate	- 84.0	77.1	6.02	6.27	1.3724	38.1
59. Fluorobenzene	- 42.2	84.7	5.42	4.90	1.4684 (15°)	38.1
60. Iodobenzene	- 31.3	188.3	4.63 (20°)	4.64	1.6200	37.9
61. Chlorobenzene	- 45.6	131.7	5.62	5.14	1.5248	37.5
62. Bromobenzene	- 30.8	155.9	5.40	5.17	1.5571 (25°)	37.5
63. Tetrahydrofuran	-108.5	66	7.58	5.84	1.4072	37.4
64. Anisole	- 37.5	153.8	4.33	4.17	1.5170	37.2
65. Phenetole	- 29.5	170.0	4.22 (20°)	4.54	1.5074	36.4
66. 1,1,1-Trichloroethane	- 30.4	74.0	7.53 (20°)	5.24	1.4379	36.2
67. 1,4-Dioxane	11.8	101.3	2.21	1.50	1.4224	36.0
68. Trichloroethylene	- 86.4	87.2	3.42 (16°)	2.7	1.4767 (21°)	35.9
69. Piperidine	- 10.5	106.4	5.8 (20°)	3.97	1.4525	35.5
70. Diphenyl ether	26.9	258.3	3.69 (20°)	3.87	1.5763 (30°)	35.3 (30°)
71. Diethyl ether	-116.3	34.6	4.34 (20°)	4.34	1.3524	34.6
72. Benzene	5.5	80.1	2.28	0.0	1.5011	34.5
73. Diisopropyl ether	- 85.5	68.3	3.88	4.20	1.3681	34.0
74. Toluene	- 95.0	110.6	2.38	1.43	1.4969	33.9
75. Di- <i>n</i> -butyl ether	- 95.2	142.2	3.08 (20°)	3.94	1.3992	33.4
76. Triethylamine	-114.7	89.5	2.42	2.90	1.4010	33.3
77. 1,3,5-Trimethyl- benzene	- 44.7	164.7	2.28 (20°)	0.0	1.4994	33.1
78. Carbon disulfide	-111.6	46.2	2.64 (20°)	0.0	1.6280	32.6

TABELE