

$$m(x) = \frac{m(x)}{M(x)} = \frac{N(x)}{NA}$$

$$p \cdot V = m \cdot R \cdot T$$

$$R = 8,314 \text{ kPa} / \text{mol} \cdot \text{K}$$

1 mol plina pri 0°C 101,3 kPa je 22,4 l

$$\bar{M} = x_i \cdot M_i = \frac{\sum n_i \cdot M_i}{\sum n_i}$$

$$p_i \cdot V_2 = n_i \cdot R \cdot T$$

$$\rho = \frac{m}{V} = \frac{p \cdot M}{R \cdot T}$$

$$\frac{p_i}{p_2} = \frac{n_i}{n_2} = x_i$$

$$y_i = \frac{V_i}{V_{\text{mezi}}} = \frac{n_i}{n_{\text{mezi}}} = x_i$$

($\cdot 10^6 \cdot 100$)
pph \rightarrow vol%
vol% (x) = $\frac{m(x)}{m_{\text{mezi}}}$
 \downarrow konc. v g/m³

1 bar = 10⁵ Pa

1 Pa = 1 N/m² = 1 kg/m s²

1 atm = 1013 mbar = 101,3 kPa

$$W_i = \frac{m_i}{m_{\text{mezi}}}$$

$V_i \cdot 100 \Rightarrow$ masni %

$y_i \cdot 100 \Rightarrow$ vol%
 \downarrow podobenost delež

$$\bar{M} = \sum \frac{m_i}{m} = \frac{m_i \cdot M_i}{m_i} = x_i M_i$$

$$p_{\text{ZMES}} = \sum p_i$$

$$SME_{\text{ZM}} = SME_{\text{mezi}} \left(1 - \frac{\text{vol}\%}{SMV_i}\right)$$

\downarrow hitrost reakt

$$\frac{\text{vol}\%_x}{\text{vol}\%_y} = \frac{n(x)}{n(y)}$$

$$n_i = \frac{m}{F}$$

$$n_i(x) \neq$$

$$\dot{m}'' = \frac{\dot{Q}'' - \dot{Q}_{\text{iz}}}{U}$$

\rightarrow hitrost prenosa

\dot{Q} - toplotni tok [kW/m²]

$$\dot{m}'' = \dot{m}''_0 (1 - e^{-kPD})$$

$$y/x = \frac{n_x}{n_y}$$

\downarrow delitev

\downarrow g/m³ \downarrow kJ/g - upornost toplote

gostota toplote

HITROST SPROSTANJA TOPLOTNE [kW]

$$\dot{Q} = \dot{m} \cdot x \cdot \Delta H_c \rightarrow \text{entalpijska sprememba [kJ/kg]}$$

$$\dot{Q} = \dot{m} \cdot x \cdot \Delta H_c$$

$$\dot{Q} = \dot{m}'' \cdot x \cdot \Delta H_c \cdot A$$

$$\dot{Q}'' = \dot{m}'' \cdot x \cdot \Delta H_c \text{ [kW/m}^2\text{]}$$

\downarrow minimalna sprememba

$$x \cdot \Delta H_c = \Delta H_{\text{ef}}$$

$$h = \frac{\dot{Q}_{\text{in}} - \dot{Q}_{\text{iz}}}{\Delta U}$$

\downarrow hitrost amoniacne gladine

$$\dot{Q} = \dot{m}_{\text{reak}} \cdot \Delta H_{\text{reak}}$$

\downarrow prostornina reaktorja

$$\dot{m}_{\text{reak}} = 0,52 \cdot A \cdot h$$

\downarrow masni tok reaktora

PREVAJANJE TOPLOTE

$$\dot{q} = h \cdot A \cdot \frac{\Delta T}{\Delta x}$$

$$\dot{q}'' = \frac{\dot{q}}{A} = h \cdot \frac{\Delta T}{\Delta x}$$

\downarrow toplotni tok [J/s = W] \downarrow koeficient topl. prevajanja \downarrow debelina materiala

$$\dot{Q} = \alpha (T - T_i)^m$$

\rightarrow časovno odvisni požari
 \rightarrow sprememba površine \rightarrow čas od nastanka

KONVEKCIJA

$$\dot{q} = h \cdot A \cdot \Delta T$$

\downarrow tok \downarrow koeficient toplotne prevajanja [W/m²°C]

SEVANJE

$$c = \lambda \cdot \nu \quad E = h \cdot \nu$$

$$\dot{q} = \epsilon \cdot A \cdot \sigma \cdot T^4$$

\downarrow emisivnost \downarrow Stefan-Boltzmann

$$\sigma = 5,67 \cdot 10^{-8} \text{ W/m}^2 \text{K}^4$$

$$\dot{q} \text{ [kW/m}^2\text{]}$$

\rightarrow toplotni sevnanje

$$\dot{q}'' = \frac{\dot{Q} \cdot x_r}{4\pi r^2}$$

\rightarrow delež toplote, ki renu iz plamena

\downarrow tok na površino v oddaljenosti r